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THE DISTRIBUTION OF GOLD AND OTHER ELEMENTS IN SOILS AND VEGETATION AT PANGLO, WESTERN AUSTRALIA

M.J. Lintern and K.M. Scott

CRC LEME OPEN FILE REPORT 44

October 1998

(CSIRO Division of Exploration Geoscience Report 129R, 1990 Second impression, 1998)









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RESEARCH ARISING FROM CSIRO/AMIRA REGOLITH GEOCHEMISTRY PROJECTS 1987-1993

In 1987, CSIRO commenced a series of multi-client research projects in regolith geology and geochemistry which were sponsored by companies in the Australian mining industry, through the Australian Mineral Industries Research Association Limited (AMIRA). The initial research program, "Exploration for concealed gold deposits, Yilgam Block, Western Australia" (1987-1993) had the aim of developing improved geological, geochemical and geophysical methods for mineral exploration that would facilitate the location of blind, buried or deeply weathered gold deposits. The program included the following projects:

P240: Laterite geochemistry for detecting concealed mineral deposits (1987-1991). Leader: Dr R.E. Smith. Its scope was development of methods for sampling and interpretation of multi-element laterite geochemistry data and application of multi-element techniques to gold and polymetallic mineral exploration in weathered terrain. The project emphasised viewing laterite geochemical dispersion patterns in their regolith-landform context at local and district scales. It was supported by 30 companies.

P241: Gold and associated elements in the regolith - dispersion processes and implications for exploration (1987-1991). Leader: Dr C.R.M. Butt.

The project investigated the distribution of ore and indicator elements in the regolith. It included studies of the mineralogical and geochemical characteristics of weathered ore deposits and wall rocks, and the chemical controls on element dispersion and concentration during regolith evolution. This was to increase the effectiveness of geochemical exploration in weathered terrain through improved understanding of weathering processes. It was supported by 26 companies.

These projects represented "an opportunity for the mineral industry to participate in a multi-disciplinary program of geoscience research aimed at developing new geological, geochemical and geophysical methods for exploration in deeply weathered Archaean terrains". This initiative recognised the unique opportunities, created by exploration and open-cut mining, to conduct detailed studies of the weathered zone, with particular emphasis on the near-surface expression of gold mineralisation. The skills of existing and specially recruited research staff from the Floreat Park and North Ryde laboratories (of the then Divisions of Minerals and Geochemistry, and Mineral Physics and Mineralogy, subsequently Exploration Geoscience and later Exploration and Mining) were integrated to form a task force with expertise in geology, mineralogy, geochemistry and geophysics. Several staff participated in more than one project. Following completion of the original projects, two continuation projects were developed.

P240A: Geochemical exploration in complex lateritic environments of the Yilgarn Craton, Western Australia (1991-1993). Leaders: Drs R.E. Smith and R.R. Anand.

The approach of viewing geochemical dispersion within a well-controlled and well-understood regolith-landform and bedrock framework at detailed and district scales continued. In this extension, focus was particularly on areas of transported cover and on more complex lateritic environments typified by the Kalgoorlie regional study. This was supported by 17 companies.

P241A: Gold and associated elements in the regolith - dispersion processes and implications for exploration. Leader: Dr. C.R.M. Butt.

The significance of gold mobilisation under present-day conditions, particularly the important relationship with pedogenic carbonate, was investigated further. In addition, attention was focussed on the recognition of primary lithologies from their weathered equivalents. This project was supported by 14 companies.

Although the confidentiality periods of the research reports have expired, the last in December 1994, they have not been made public until now. Publishing the reports through the CRC LEME Report Series is seen as an appropriate means of doing this. By making available the results of the research and the authors' interpretations, it is hoped that the reports will provide source data for future research and be useful for teaching. CRC LEME acknowledges the Australian Mineral Industries Research Association and CSIRO Division of Exploration and Mining for authorisation to publish these reports. It is intended that publication of the reports will be a substantial additional factor in transferring technology to aid the Australian Mineral Industry.

This report (CRC LEME Open File Report 44) is a Second impression (second printing) of CSIRO, Division of Exploration Geoscience Restricted Report 129R, first issued in 1990, which formed part of the CSIRO/AMIRA Project P241.

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Preface

The CSIRO-AMIRA project "Exploration for Concealed Gold Deposits, Yilgarn Block, Western Australia" has as its overall aim the development of improved geological, geochemical and geophysical methods for mineral exploration that will facilitate the location of blind, concealed or deeply weathered gold deposits. This Report presents results of research conducted as part of Module 2 of this project (AMIRA Project 241): "Gold and Associated Elements in the Regolith - Dispersion Processes and Implications for Exploration".

The objectives of this module are, inter alia, to suggest improvements in techniques of sampling and data interpretation for gold exploration, and to increase knowledge of the properties and genesis of the regolith. report describes the geochemistry of soils and vegetation of the Panglo gold deposit north of Kalgoorlie. The results contribute strongly to the objectives of the module by (i) providing further evidence (already demonstrated at Mt. Hope) of the strong association between Au and the alkaline earth metals, (ii) examining the distribution of other geochemical components of the regolith, (iii) sampling, analysing and interpreting geochemical results of vegetation and (iv) studying electron micrographs to elucidate the role of pedogenic carbonates in near-surface weathering processes. The work is of major significance to gold exploration as it has demonstrated that gold is preferentially concentrated in the soil carbonate horizon, even when the mineralization is concealed by leached saprolites and barren transported overburden. The research has also established that the phenomenon is of regional significance, having previously been documented at Mt. Hope, 250 km to the south-west. The principal conclusion is that it is essential to include the carbonate horizon, where present, in any soil sampling programme for gold exploration.

C.R.M. Butt,
Project Leader.
October, 1990.

Abstract

Shallow auger, surface, profile, trench, grab and vegetation samples were taken along two traverses at the Panglo gold deposit about 30 kilometres NNW of Kalgoorlie, Western Australia. Landscape and soil features were photographed, described and used to interpret geochemical information. The relationship between the geochemistry of soils and vegetation was explored with particular reference to Au. Detailed examination of weathering surface material was performed using SEM (scanning electron microscopy) techniques.

The results establish a widespread and strong association between Au and pedogenic carbonate. The most significant result was the presence of Au (up to several hundred ppb) in pedogenic carbonate occurring in transported overburden above mineralization. There was also a weak association between Au found in vegetation and that found in the soils. Other results indicated the strong association between landscape features, geochemistry and vegetation type. SEM studies revealed the importance of calcite in near-surface weathering processes.

The study demonstrates the importance of locating and sampling the pedogenic carbonate horizon if present. Augering and near-surface sampling appear to be the most successful exploration techniques whereas trench sampling may falsely locate or overlook the soil anomaly altogether. Vegetation sampling alone cannot be used to locate mineralization.

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The Distribution of Gold and Other Elements of Soils and Vegetation at Panglo, Western Australia.

M.J. Lintern and K.M. Scott.

1. Introduction.

The Panglo gold deposit is situated approximately 30km kilometres NNW of Kalgoorlie between the Kalgoorlie-Leonora railway and adjacent Kalgoorlie-Meekatharra highway (Figure 1).

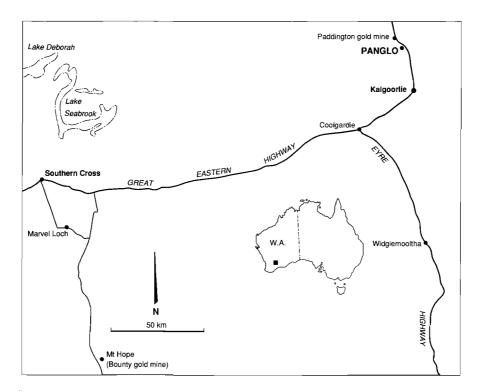


Figure 1: Map showing location of the Panglo deposit and the Bounty gold mine.

It was discovered in late 1986 by Pancontinental Mining Company Limited following routine soil sampling south of the Paddington mine. In 1989, a large test pit was excavated as part of mining evaluation procedures. The deposit comprises a mineable resource of 1 mt at 3.6 gpt and is due to start production shortly.

At the commencement of the study, the surface environment was relatively free from potential contamination and provided a favourable site for a detailed examination of the distribution of gold in soils.

The presence, at Panglo, of surface pedogenic carbonate is of particular importance. The studies at Mt Hope, situated over 250km to the south-west of Panglo (Figure 1), had indicated a strong association between soil carbonate and Au (Lintern, 1989). Demonstration of the same association in a different area has considerable implications for exploration as it overwhelmingly suggests that it is a regional phenomenon.

2. Scope of study.

The present study at Panglo includes a brief description of the relationships between the landscape, vegetation, soils and geology. A geochemical approach is used to relate these different aspects of the deposit to the weathering process. Particular attention is paid to soil geochemistry and mineralogy, and biogeochemistry on two traverses across supergene mineralization.

3. Site Description.

The Panglo deposit is situated to the west of an area of some salt scalding, a prominent surface feature that is clearly defined in aerial photographs by its light colour and diminished vegetation (Figure 2). The deposit itself is situated partly within, but mainly adjacent to, a broad, poorly-defined, highly saline ephemeral drainage channel. Towards the southern end of the mineralized zone, at 3700N, the drainage narrows to a relatively flat channel, approximately 100m wide confined by low steep slopes. The gradient of the drainage channel decreases from north to south (Figure 2).

There is little outcrop over mineralization but mineralized lithorelics of various sizes (ranging up to several centimetres) occur in a matrix of heavy clays within a few centimetres of the surface. The test pit reveals that bleached saprolitic clays are a prominent feature of the weathering profile. The depth of weathering is in excess of 60m. Economic grades of

mineralization occur as a supergene deposit at a depth of about 40m, hosted by weathered shales and mafic volcanics within an 800m wide shear zone. Drilling indicates that the deposit is orientated SSW-NNE. Bedrock sources of gold mineralization have not been determined to date.

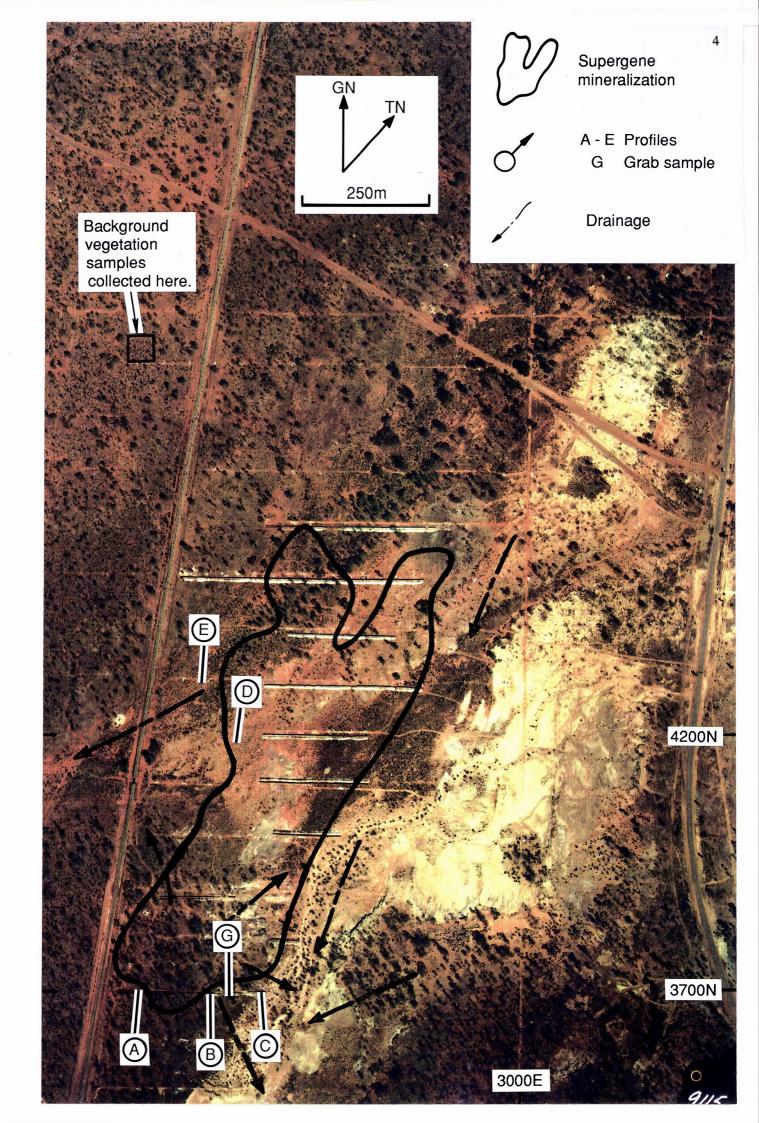
To the west and south-west of mineralization, a palaeodrainage of presumed Tertiary age has been defined (R. Howard, pers comm). It partially encroaches over the southern limit of mineralization, where its presence is indicated near the surface by loosely cemented lateritic gravels.

Soil development and character over the deposit and adjacent areas is broadly determined by geomorphology but locally determined by geology. In the drainage channels and broad flats, clay-rich soils dominate; on higher ground, soils are usually gravelly. Soils are often thin (generally <0.5m) and locally absent over outcropping barren undifferentiated sediments and siltstone/shale units to the west of mineralization. Generally, over mineralization, soils can be grouped into two classes: (i) the gravelly soils and (ii) the clay-rich soils.

(i) The gravelly soils occur in higher areas of the landscape and include those over the known southern limits of mineralization. They include those found in the palaeochannel. The soils consist of an organic-rich surface horizon containing Fe-rich gravels in a matrix of sandy material. Beneath this, the soil is characterized by unconsolidated, Fe-rich, rounded to subrounded ferruginous nodules varying in size from a few millimetres to several centimetres in diameter in a sandy to loamy matrix. Saprolite, at or near the surface in certain areas, limits soil depth to a few centimetres. Carbonate development occurs near the surface and continues to several tens of centimetres. Carbonate occurs as a) friable fragments and thin veneers in unconsolidated Fe-rich material and b) as coatings, up to several millimetres thick, on consolidated near-surface saprolite.

Figure 2: Aerial photograph with plan of the study area.

Air photograph probably taken by Kevron Air Surveys 15th Dec 1986, Photo 9115, published with permission of Goldfields Exploration Pty Ltd.

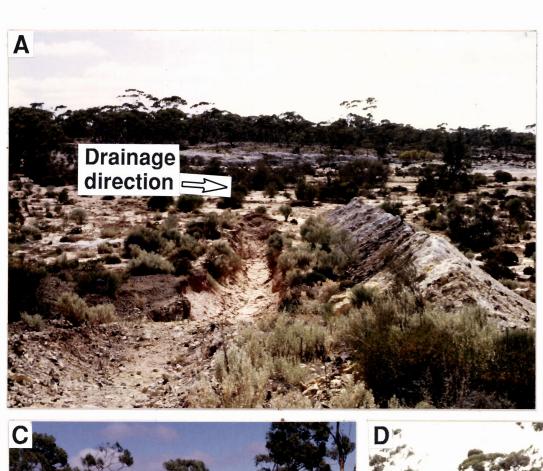


(ii) The clay-rich soils are dominant in the broad valley flats of the northern and central sections over the mineralized zone. They consist of a veneer of sand-rich material overlying pale grey to yellow fine homogeneous clays. Blocks of weathered rock are common within the soil profile and are either pale (clay-rich) or red (Fe-rich) coloured. Carbonate occurs within the profile as friable aggregates and veneers.

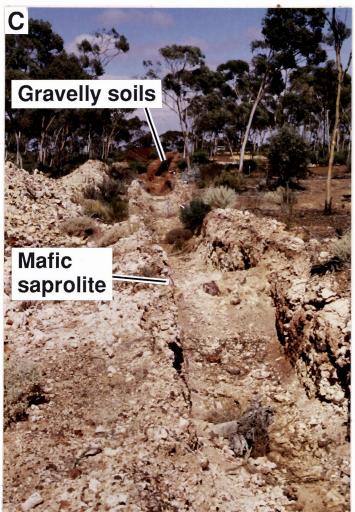
The diversity and abundance of the vegetation is determined by soil characteristics. Salt- and drought-tolerant species e.g. Maeriana spp. and Atriplex spp. (<0.5m in height) with minor Eremophila spp. (up to 2m) dominate the clay-rich soils of the broad valley floor to form an open heathland community. Where soils are thin, the vegetation is reduced to sporadic occurrences and the total biomass is low. Tall Eucalyptus spp. trees (up to 20m) with an under storey of Eremophila spp. (1-2m) form a medium dense woodland community over the gravelly soils. There is a sharp transition between the two vegetation communities reflecting the different soil types, although individual examples of plant species occur sporadically in both communities.

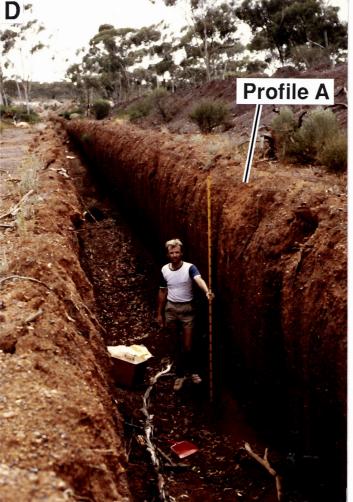
Figure 3: Photographs of the study area (overleaf).

- a) Traverse 3700N looking east across the drainage channel.
- b) Traverse 4200N looking west from 3000E across the drainage flat.
- c) Traverse 3700N looking west from the weathered rock outcrop.
- d) Traverse 3700N looking east from Profile A.
- e) Large boulder of weathered rock.
- f) Profile A.
- g) Profile B.
- h) Profile C.

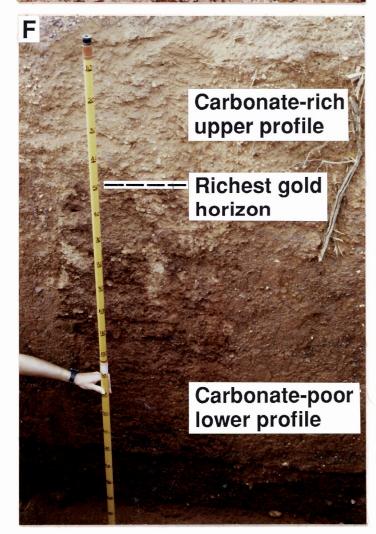


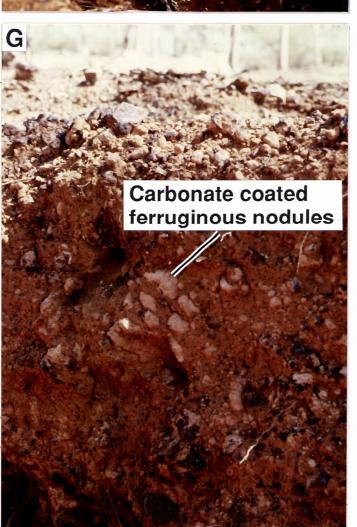














4. Methods.

4.1. Sample collection.

Soil and vegetation samples were collected during winter 1988 and 1989. Pancontinental Ltd. provided surface soil, auger and trench samples for line 3700N and some surface soil samples from line 4200N (Table 1).

Sample Type	Northing sampled	Collected by
Profile	3700 4200 4300	CSIRO
Surface	3700 4200	CSIRO, Pancontinental
Auger	3700	Pancontinental
Trench	3700	Pancontinental
Vegetation	3700 4200 + others	CSIRO
Special,		CSIRO

Table 1: Sample types, northings and collection summary.

Sample locations are shown in Figure 2. Lines 3700N and 4200N were chosen as the main sampling sites as they had the following attributes:

- (i) Access was provided by trenching .
- (ii) They occurred above mineralization.
- (iii) Several lithologies and landscape features were represented.
- (iv) Different vegetation communities were represented.

4.1.1. Profile.

Profiles were channel sampled in existing trenches and pits. Profiles were first cleaned to remove foreign material and photographed before sampling. Samples were obtained sequentially, from top to bottom, after benching of the profile using a geological hammer and rock chisels (where necessary). Samples were collected into a plastic pan before transferring to polyethylene bags. The benching technique ensured minimal crosscontamination from material higher up in the profile. Composite samples

were generally collected over 10cm intervals but were varied according to the type of soil material being sampled.

4.1.2. Surface.

Decaying vegetation and/or foreign material if present were removed from the top 5cm and samples (about 1kg) collected by hand auger to 30cm depth, or refusal, along 3700N, and by hand from the top 10cm along 4200N. Subcropping weathered rock samples were taken from 4200N from about 30cm depth.

4.1.3. Power auger.

Composite auger sampling over 0 to 1.2m (one rod) was carried out using a vehicle-mounted power auger on 3700N.

4.1.4. Trench.

Samples were collected at a set height (about 0.5m or "knee-height") above the base of the trench. The depth of the trench was dependent upon penetration by the earth-moving equipment is thus of considerable importance. Over unconsolidated transported material, the trench is over 2m deep but over the weathered rock penetration was restricted to a few centimetres. The depth of the "knee-height" sampling method relative to the surface, therefore, resulted in different horizons of the soil profile being sampled. The north face of the trench at 3700N was sampled. Profiles in the trench (A, B and C) were sampled as described in 4.1.1 above.

4.1.5. Vegetation.

Vegetation sampling was restricted, where possible, to new growth. Samples were taken of leaves and some stems and broken off by hand or using secateurs. Vegetation samples were collected in plastic bags which were left open, to facilitate drying, before transfer to the laboratory.

4.1.6. Special samples.

Two grab samples were collected from locations near to Profile B for detailed geochemical studies.

4.2. Sample preparation.

Profile, near-surface and auger samples were dried, weighed, homogenized and split on a plastic sheet. Half the sample was bagged for reference and underwent no further treatment. Some near-surface material (500g subsample) underwent no further preparation prior to Au analysis by cyanide leaching.

Profile samples containing coarser material were crushed prior to splitting using a hydraulic press mounted with zirconia plates. Other samples containing coarser material (greater than 25 mm) were reduced in size using a Mn-steel jaw crusher. Approximately 50g of the split was incrementally extracted from the sample and ground to a nominal minus 75 μ m in an agate ring mill. Self-supporting pressed discs were prepared for XRF analysis. Sub-samples were extracted for X-ray diffraction (XRD), inductively coupled plasma (ICP) and atomic absorption spectroscopy (AAS) analysis.

Sub-samples of carbonate and saprolite material from grab samples were obtained by percussion using a hand drill. Sub-samples were selected for polished sections and examination by scanning electron microscopy (SEM).

Vegetation samples, approximately 30g in weight, were placed in a plastic sieve and rinsed to remove soil material. The sample was then dried at 105°C for at least 24 hours to prevent smearing during milling.

The vegetation samples were passed through a four-bladed cross-beater mill twice - firstly without, and secondly with, a 1 mm mesh screen in place to ensure a suitably homogenized and macerated sample for instrumental analysis. Samples required no further preparation for instrumental neutron activation analysis (INAA) but required stepwise ashing up to 600°C for 24 hours prior to other analyses by ICP and graphite furnace AAS. Samples were weighed before and after ashing so that data standardization could be made with the un-ashed samples sent for INAA.

4.3. Sample analysis.

Several analytical methods were used depending upon sample size, detection limits and instrument availability. In consequence, some elements were recorded by different methods. Standards were submitted with samples for each analytical batch to ensure results were consistent throughout the study. Satisfactory agreement was also found on selected elements using different analytical techniques on aliquots of the same sample. Only the most reliable data are reported upon. The instruments and methods used are listed below:

XRF - Pressed discs, prepared by the method of Norrish and Chappell (1977) were analysed for minor and trace elements using a Phillips PW 1200C XRF. Major elements were analysed using fused discs.

ICP - For soils, approximately 250 mg of sample was accurately weighed out and mixed with 1 g of lithium metaborate. The mix was fused at 1000°C in a furnace for one hour and the residue dissolved in 1N nitric acid. The solutions were analysed on either a Labtem or a Hilger E-100 ICP. For vegetation, the ashed sample (dry weight of 10g) was digested in aqua regia, diluted and analysed.

OES - 50mg of sample and buffer (LiF) was arced in a graphite anode and analysed by a Hilger Emission Spectrometer. Results are semi-quantitative as no matrix correction was applied.

ICP-MS - Soils were shaken with a solution of 0.2% KCN saturated with Ca(OH)₂ in the ratio of 1:2. The bottle, containing the sample and solution, was capped and agitated for a minimum of 24 hours. After standing to allow the clay particles to settle, at least 4mL of solution was collected by pipette and analysed directly for Au after 1:10 dilution at Sheen Laboratories, WA.

ICP-OES - Following mixed acid digestion samples were analysed by Genalysis, WA.

AAS - Samples were either digested in cyanide (CN) as for ICP-MS or by aqua regia (AR) and analysed for Au. The graphite furnace attachment was used for all samples.

INAA - Vegetation samples were pressed into discs, and soil and rock samples were accurately weighed (30g) in plastic containers prior to being sent for analysis at Becquerel Laboratories, NSW. Vegetation analysis using INAA was subject to considerable interference of Na on Au, resulting in unacceptably high detection limits. The problem was not resolved and an alternative aqua regia sample dissolution / graphite furnace AAS procedure adopted.

XRD - Homogenized powdered samples were packed and irradiated with CuKoK radiation at 40kV using a Phillips PW1050 Diffractometer.

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4.3.1. Profile.
```

ICP - SiO₂, TiO₂, Al₂O₃, Na₂O, Fe₂O₃, MgO, CaO, P₂O₅, Ba, Be, Co, and Cr.

XRF - S, Ce, Cu, Ga, Ge, Mn, Mo, Nb, Ni, Pb, Rb, Sr, V, Y, Zn and Zr.

INAA - As, Au (Profiles A, B and C), Br, Sb and W.

AAS - Au (CN, Profiles D and E)

XRD - On selected samples.

4.3.2. Surface.

ICP - Na_2O , Co, Cr, Cu, Ni, Pb (4200N soils), Sc, Sr, V, Y and Zn.

XRF - SiO_2 , TiO_2 , Al_2O_3 , Fe_2O_3 , S, MgO, CaO, P_2O_5 , K_2O , As, Ba, Pb (4200N rocks) and Zr.

INAA - Au (4200N), Sb and W.

OES - Ag, B, Bi, Ga, Ge, Mo, Sn and Tl.

AAS - Au (CN, uncrushed soils)

XRD - On selected samples.

4.3.3. Auger.

ICP - SiO2, TiO2, Al2O3, Na2O, Fe2O3, MgO, CaO, Co and Cr.

XRF - S, Ba, Ce, Cu, Mn, Ni, Nb, Pb, Rb, Sr, V, Y, Zn and Zr.

AAS - Au (AR).

4.3.4. Trench.

AAS - Au (AR).

ICP-OES - As.

4.3.5. Vegetation.

ICP - Si, Al, Mg, Ca, Ti, Na, P, Ba, Be, Co, Cr, Cu, Mn, Ni, V
 and Zr.

INAA - As, Br and Sb.

AAS - Au (AR).

4.3.6. Special.

ICP - SiO_2 , Al_2O_3 , Fe_2O_3 , MgO, CaO, Na_2O , TiO_2 , P_2O_5 , Ba, Be, Co, Cu, Mn, Ni, V and Zr.

XRF - S, Ce, Ga, Ge, Mo, Nb, Pb, Rb, Sr, Y and Zn.

AAS - Au (CN).

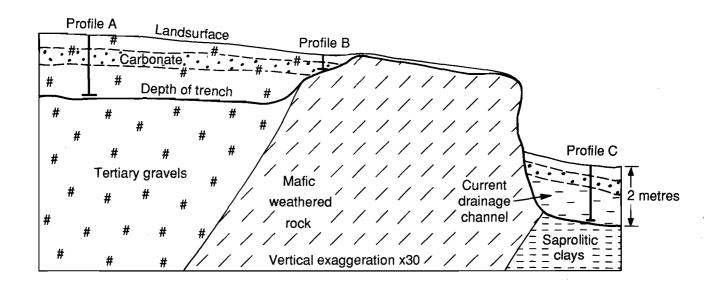


Figure 4: Section through part of the trench on 3700N showing position of profiles and depth of trench.

5. Results and Discussion.

All the profile, near-surface, auger, trench and vegetation geochemical results are plotted and tabulated in the Appendices. The most significant data are discussed below.

5.1. Profile samples.

Profiles were sampled in the three principal geomorphological sites of the trench on traverse 3700N. These were in soils developed on (i) the transported lateritic gravels in the Tertiary palaeochannel (profile A), (ii) the weathered mafic rock (B) and (iii) the current drainage channel (C) (Figure 4). Profiles A and B were taken directly over the southern limits of the known mineralization. An additional two shallow profiles (D and E) were sampled near the central section of supergene mineralization on traverses 4200N and 4300N.

5.1.1. Profile A (2278E, 3700N).

This profile is developed upon transported material associated with a Tertiary palaeochannel. The overburden consists primarily of a sediment of pisoliths and nodules which include fragments of amphibolite. Drilling indicates the overburden to be 10m thick and is comprised of gravels overlying mottled clays. The overburden is situated on clay-rich saprolite derived from mafic volcanics several tens of metres thick. There is some Au enrichment at the boundary between the clay-rich mottles in the palaeochannel and the clay-rich saprolite. Economic grades of Au occur as a supergene deposit within the saprolite at about 40m.

The principal horizons in the profile are as follows:

0-10cm. Red clay-rich sandy loam with ferruginous fragments (up to 20mm). Carbonate present as minor aggregates and coatings. Some organic material. 10-65cm. Sandy loam with gravels. Large amounts of carbonate present as coatings and aggregates progressively increasing then decreasing downwards. 65cm-2.2m. Sandy loam with gravels. Decreasing amounts of carbonate downwards with some present as pendants and lining old root channels. Little, if any, carbonate present in lower parts of the horizon.

The principal minerals present in the profile are hematite, goethite (in the gravels) and quartz. Calcite increases to a maximum at about 0.5m and then decreases. Little XRD-detectable calcite appears below about 1m. Kaolinite, although present, is not abundant. Some halite is present towards the base of the profile.

Near-surface material is rich in Ti, Fe, Mg and Ca, which are probably associated with the coarser material. Below this, Ca concentrations increase and are associated with the calcite. There appears to be an antipathetic relationship between Ca and Fe (Figure 5b) suggesting that calcite has been introduced at a later stage and has diluted the concentration of Fe minerals.

There is a very strong association between Au and the alkaline earth metals Ca, Mg and Sr (Figure 5a) throughout the profile. The lowest concentrations occur at the base of the soil profile (Figure 5b; Appendix 1). Peak concentrations of Au (0.26 ppm) and the alkaline earths (e.g. 16% CaO) occur at 45cm depth.

At the surface, As, Ce, Cr, Cu, Mn, Pb, Rb, Sb, Sr, V, W, Y, Zn and Zr amounts are high and are probably associated with the Mg, Ca, Fe and zircon minerals. Throughout the profile, Fe oxides appear to control the distributions of Ti, Al, As, Cr, Ga, Pb, Sb and V, and, to a lesser extent, with Y and Zr between 0.6m and 1.2m (Appendix 1). Concentrations of these elements are comparatively poor (especially Fe) when Ca contents are high and are consistent with the proposed dilution effect caused by the introduction of carbonate (Figure 5b).

Sulphur, Br and Na concentrations increase with depth (Appendix 1) particularly below about 1.6m. This may be due to continued evaporation of brackish drainage water that has flowed into the trench. Alternatively, the high salt content may be related to the very saline groundwater which occurs at about 10m depth.

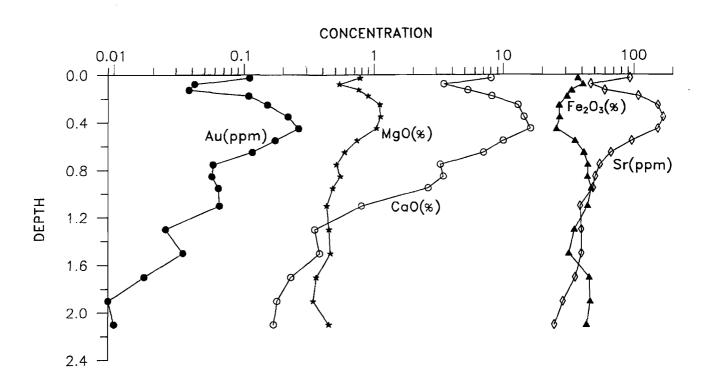


Figure 5a: Distribution of CaO, MgO, Sr and Au in Profile A.

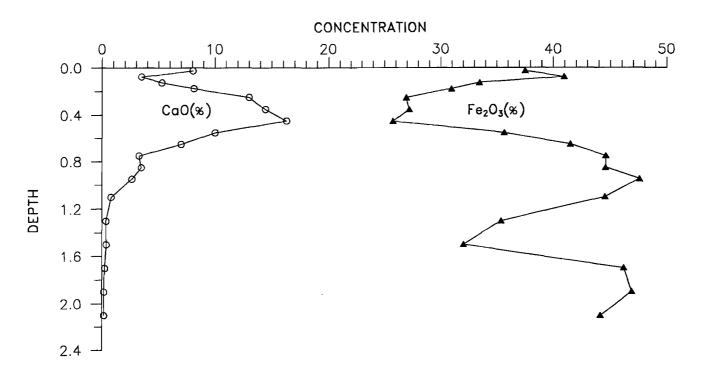


Figure 5b: Distribution of Fe_2O_3 and CaO in Profile A.

5.1.2. Profile B (2375E, 3700N).

This part of the trench is characterized by the presence of outcropping weathered mafic volcanics. The profile consists of a thin, poorly developed soil overlying ferruginous saprolitic fragments of Fe-rich nodules partially cemented by carbonate. Indurated, weathered mafic volcanics continue with depth. The zones of Au mineralization are as for Profile A.

The principal horizons in the profile are as follows:

0-5cm. Thin sandy soil with large quantities of ferruginous lithorelics. Carbonate present as coatings, aggregates and separated veneers. Some organic material is present.

5-80cm. Sub-rounded and angular cobbles of ferruginous lithorelics whose size and degree of induration increases with depth to become massive boulders (>10cm) of weathered rock at the bottom of the profile. Carbonate present as matrix, aggregates and coatings throughout the profile but becomes less abundant with depth.

The principal minerals present in the profile are hematite and goethite with some calcite and quartz. Hematite and goethite tend to increase in concentration down the profile while calcite decreases. The profile is kaolinite poor.

The relative influence of weathered mafic volcanic rocks and introduced carbonate are reflected in the subtle antipathetic relationship between Fe and Ca concentrations (Figure 6). Close examination of the ferruginous material indicates that any Ca present occurs as coatings or in cracks (see Section 5.6.3). The emplacement of calcite within the matrix and cracks is consistent with the late stage introduction noted in Profile A.

The distribution of Au shows a tendency to decrease down the profile and, on this evidence alone, is seen to be weakly related to the carbonate distribution (Figure 6). Maximum concentrations of Au (about 0.13 ppm) occur at about 0.5m and occur with CaO concentrations of about 9%. There is, however, a very strong association between Ca and Au. This is demonstrated by close examination of the special samples taken from this area: see section 5.6.3.

Interpretation of minor and trace element distributions in this profile is primarily determined by the distinction made between the Fe oxide and the calcite (Ca) phases and less so with Na and Si. Thus sympathetic associations are found with (i) As, Cu and V, and Fe, (ii) Mg, Pb and Sr, and Ca, (iii) Br and Na and (iv) Ti and Zr, and Si (Appendix 2). Many minor and trace elements, e.g. Ce, Nb and Rb show no distinct trends in distribution in the soil profile (Appendix 2).

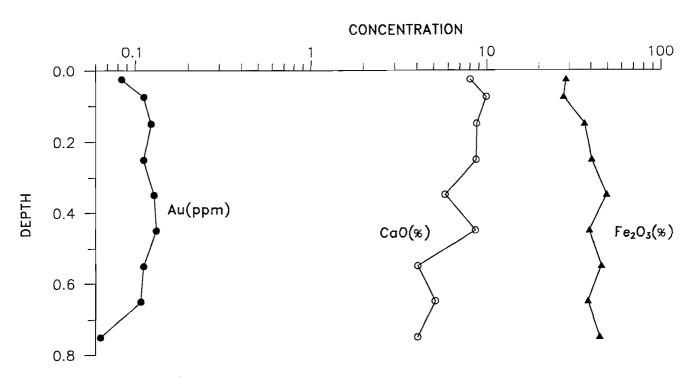


Figure 6: Geochemistry of Fe₂O₃, CaO and Au in Profile B.

5.1.3. Profile C (2505E, 3700N).

This profile is developed in the current drainage channel above unmineralized shales. It is dominated by considerable quantities of transported clay minerals and quartz. The boundary of the transported horizon which is near the base of the trench, is marked by the presence of clay-rich saprolite derived from underlying shales.

The principal horizons are as follows:

0-20cm. Clay-rich sand with appreciable quantities of ferruginous, siliceous and carbonate fragments and little organic material present.

20-60cm. Clay-rich sand with some carbonate and other coarse sized fragments present.

60-90cm. Mainly clay-rich sand with some present as friable aggregates. Minor quantities of coarse sized material present.

90-1.6m. Horizon dominated by coarse angular quartz with some finer subrounded quartz fragments. Some coarse lithorelics have similarities with the mafic volcanics found in Profile B.

1.6-1.8m. Yellow, grey and red clay-rich sand with some layered mica-rich nodules present. Little, if any, coarse quartz or ferruginous fragments present.

Calcite is restricted, in the upper part of the profile, to occasional small friable aggregates whose structure suggests they are derived from the calcite coating the adjacent mafic volcanics e.g. those found in Profile B. Goethite and hematite are associated with ferruginous fragments throughout the profile. The fragments present in the upper part of the profile also show similarities with the mafic volcanics of Profile B. quantities of quartz are present as angular detrital fragments (up to 5mm in length) that are especially in the lower parts of the profile immediately above the clay-rich saprolite. Clay minerals are present as kaolinite; muscovite increases in abundance at the base of the profile. Calcium and Mg decrease markedly down the profile and are associated with calcite and fragments of mafic volcanic rocks. Aluminium concentrations are closely related to the distribution of the clay minerals. contents are high throughout the profile but peak near the base where the volume of quartz fragments increases. Iron concentrations are low compared with Profiles A and B.

Gold distribution is predominantly associated with the fine (<0.2mm) fraction (Appendix 7). There may be a weak association between Au and Ca (Au = 0.035 ppm, CaO = 0.82 %) in the upper part of the profile (Figure 7) but the highest Au concentration (0.075 ppm) occurs in the fine fraction associated with kaolinite and muscovite.

Sodium, Br, Ga, Mo, Nb, Rb, Sb, Sr and Zr distributions are closely associated with Al. Their concentrations tend to be high at about 0.5m, low at 1.3m, before peaking in concentration at the bottom of the profile (1.7m). The elements are probably associated with cation-exchange or substitution sites on muscovite. Sulphur, Co, Mn, Ni and Zn abundances decrease down the profile and may be associated with more than one soil phase e.g. calcite, barite or organic matter.

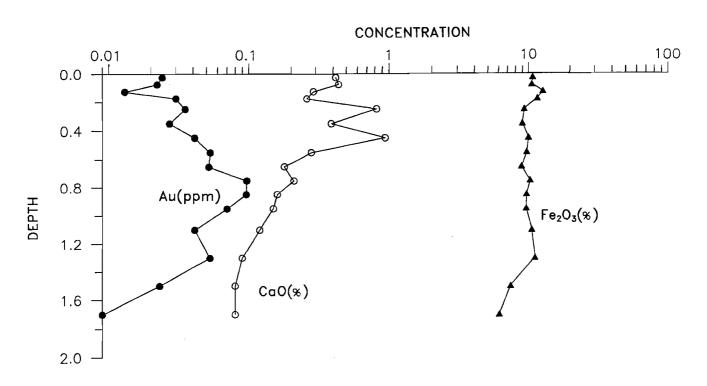


Figure 7: Distribution of Fe₂O₃, CaO and Au in Profile C.

5.1.4. Profile D (2420E, 4200N).

This soil profile, at the site of RC drill hole PSRC 231, has developed above ultramafic rocks (Scott, 1989b). The near-surface material is considered to have a strong transported component (Scott and Dickson, 1989). Beneath this horizon, mottled and clay-rich saprolite predominate for several tens of metres. Supergene mineralization occurs at about 40m. The principal horizons in the profile are as follows:

0-10cm. Clay-rich sand present with some organic litter.

10-50cm. Becoming increasingly rich in clay minerals and poor in quartz. Carbonate development increases with depth. Ferruginous fragments present throughout the profile. Large boulders of saprolite present in adjacent trench (c.f. Section 5.2.)

Quartz is the major mineral present (especially in the top 10cm) with kaolinite, hematite, goethite, calcite, talc and minor plagioclase(?). Quartz concentrations decrease, and carbonate and hematite concentrations increase with depth.

Silicon contents are closely related to the abundance of quartz. These are greatest near the surface and probably indicate transported material (Scott and Dickson, 1989). Iron is associated with the occurrence of hematite, Al with kaolinite, and Ca and Mg with calcite; the high Mg/Ca ratios (>1) suggest the calcite is present as a high-Mg variety, although some of the Mg will be present as talc. No dolomite was found.

The highest Au concentration (0.210 ppm) appears to be strongly related to the high Fe concentration (17.4 %) at the bottom of the profile (0.45m) but there may also be a weak association between Au and calcite (Figure 8).

Rubidium (especially), Mn, Ni, P, Zn and Zr concentrations appear to be related to Al concentrations (Appendix 1.4) although some of the Zn and Ni could occur with talc (c.f. Scott 1989b). Chromium and V distributions are associated with Fe (Appendix 3; cf also Scott (1989b)). Strontium, S, Na and Ba concentrations are closely related to those of Ca.

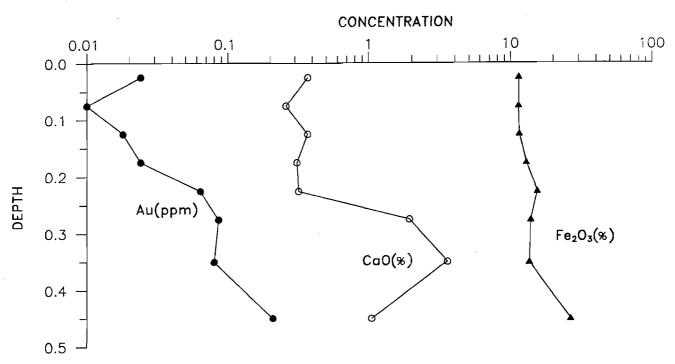


Figure 8: Distribution of Fe_2O_3 , CaO and Au in Profile D.

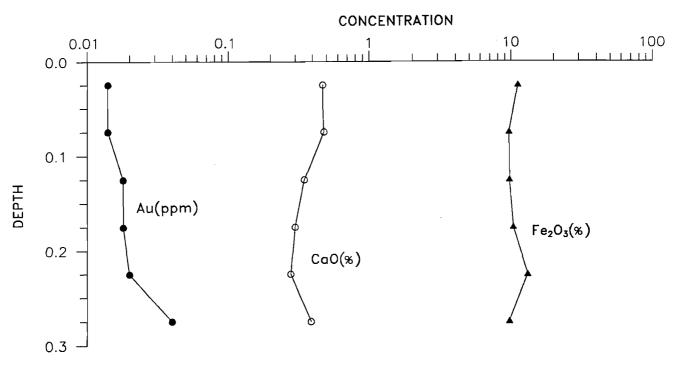


Figure 9: Distribution of Fe₂O₃, CaO and Au in Profile E.

5.1.5. Profile E (2340E, 4300N).

This shallow profile (30cm) is developed above ultramafic volcanic rocks 40m west of PSRC 227. It is similar in appearance to Profile D and consists of loamy sand overlying clay-rich sand.

The profile is rich in quartz, kaolinite and hematite with some talc and goethite, and trace amounts of anatase. No carbonates were detected by XRD in the top 30cm. However, calcite occurs in the 0-1m composite sample in PSRC 227 (Scott, 1990) and field observations in the adjacent trench suggest that carbonate is probably present below about 40cm i.e. beneath the depth of sampling.

Major element distributions are similar to those for Profile D. Silicon maxima concentrations are associated with quartz (although some Si will be in kaolinite and other silicates), Fe with hematite and most Al with kaolinite. Calcium concentrations are relatively poor.

The abundance of Au increases down the profile and does not appear to be associated with any particular mineral phase; the sharp increase in Au at the base of the profile, however, may be related to the top of the carbonate horizon.

Iron concentrations are strongly associated with As, Nb, Pb, V, Ba and Cr. Some cations (e.g. Ca, Ga, Mn, Pb, Rb), are strongly associated with the distribution of Al, where they may be present as interlayer ions in kaolinite, although most of the Mg, Cu, Ni, Zn will be associated with the occurrence of talc (Appendix 1.5; cf also Scott 1990).

5.2. Surface samples.

Surficial soil samples were collected over mineralization, adjacent to the trench at 4200N; rock samples from the trench were also collected. Analytical results are summarised in Appendix 2. The traverse crosses shales, ultramafic and mafic rocks. The soils are considered to have developed in shallow transported material (Scott and Dickson, 1989). Supergene mineralization occurs at about 40m.

There is a strong association between Au and Mg, Ca and Sr (Figure 10). The association is remarkable as it occurs, for Au and Ca, over almost seven orders of magnitude. Most of the Ca appears to be present as secondary low-Mg calcite (from probe studies) which occurs as friable aggregates (<2mm in diameter), coatings on ferruginous fragments and in the -63μ m size fraction. Some of the Mg, however, is probably distributed between vermiculite-chlorite and talc which are present in considerable quantities in some samples. Strontium is commonly associated with calcite at Panglo (c.f. the profiles).

There is a strong association between Fe and As, Cr, Sb, V and, probably W, reflecting the incorporation of these elements into secondary Fe oxides (Appendix 2; see also Scott 1990). These associations are particularly well developed in the Fe-rich rocks.

The surface rock samples contain little calcite; any Ca is likely to be present as gypsum as implied by the association of Ca and S (Appendix 2). There is no association between Au and Ca in the rock samples along line 4200N.

The presence of large amounts of plagioclase in soils over various lithologies, although there is little or none in the rocks, is further evidence that these soils have developed on transported material (Appendix 2; see also Scott and Dickson, 1989).

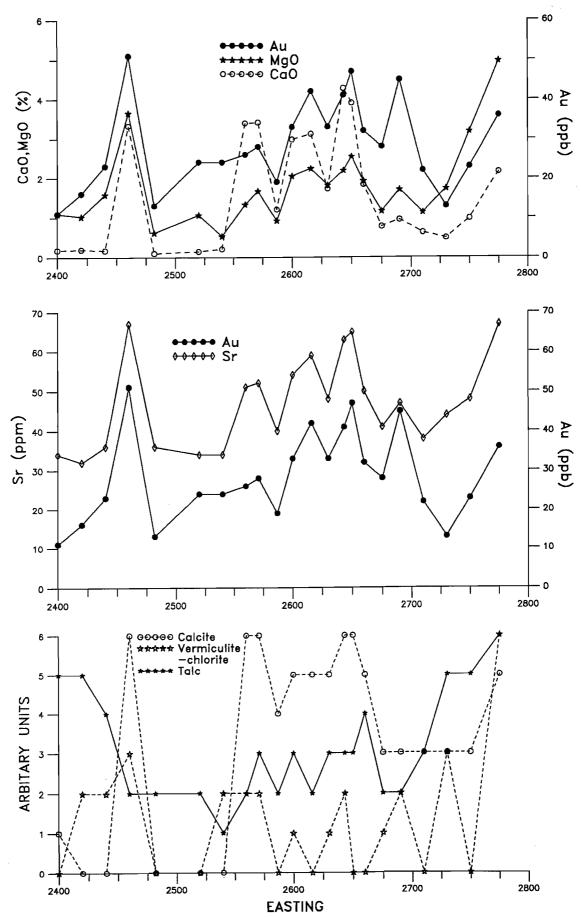


Figure 10: Distributions of a) Au, CaO, and MgO b) Au and Sr, and c) talc, calcite and vermiculite-chlorite in surficial soils at 4200N.

A more extensive suite of surficial soils from traverse 4200N and 3700N were specifically analysed for Au; the samples were not crushed or pulverized before digestion with cyanide solution. The procedure has similarities with the bulk leach extractable gold or BLEG technique but uses a smaller sample volume (500g compared with 1000g).

Results are compared with pulverized samples after fire assay fusion (Figure 11). Whereas it is clear that more Au was found by fire assay, the distributions are similar by the two methods. The ratio of background to anomaly is also similar being about 1:10 for 3700N and 1:5 for 4200N. This indicates that a significant quantity of Au (up to 100%) in these soils is freely available to chemical dissolution. It also implies that a large proportion of Au is associated with the fine fraction and not encapsulated within larger impervious soil aggregates or saprolitic fragments.

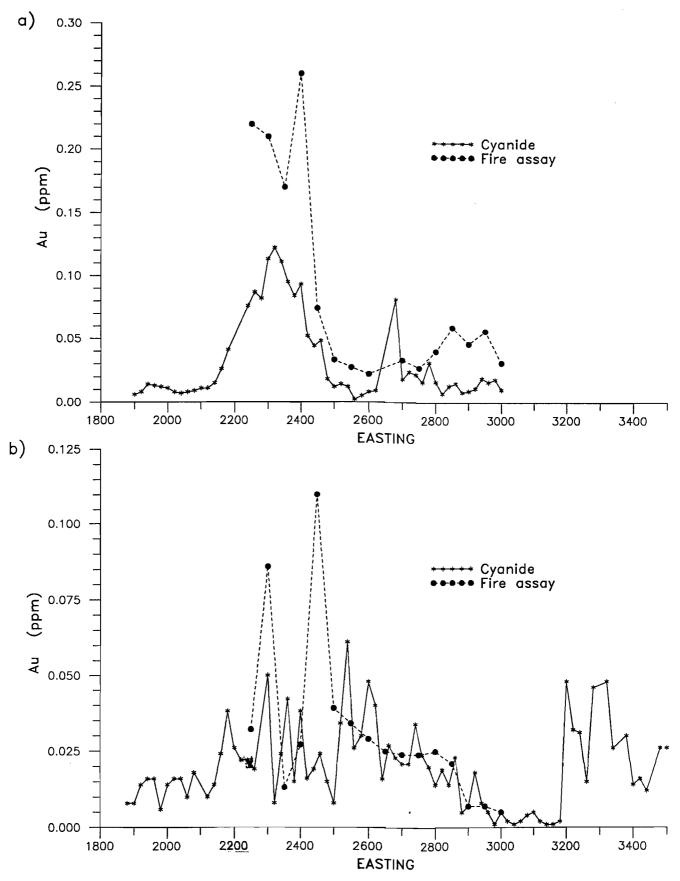


Figure 11: Comparison of fire assay-extractable Au from pulverized soil samples and cyanide-extractable Au in uncrushed soil samples from traverses a) 3700N and b) 4200N.

5.3. Auger samples.

The auger traverse on 3700N, drilled by Pancontinental Mining Ltd., sampled soils developed on the transported lateritic gravels, outcropping weathered mafic rock and the current saline drainage channel. The results show a strong association between sample type and composition (Appendix 3). The principal features of the distributions are discussed below:

- (i) Au concentrations gradually increase, west to east, from the transported lateritic gravels (0.020 ppm) to the weathered mafic rock (0.170 ppm), and then decrease into the channel soils (0.020 ppm). One sample in the channel is particularly rich in Au (0.106 ppm). The Au here may be associated with transported fragments from the adjacent hill (Figure 12a; and also Profile C);
- (ii) Silicon, Na and S, Rb, Zn and Zr contents are greatest in the saline clay-rich channel soils (Figure 12b; Appendix 3);
- (iii) High Fe, Ca, Mg, Co, Ni, Cr, V, Mn, Cu and Sr contents generally occur in the lateritic gravels, in which they are associated with ferruginous saprolitic or carbonate fragments and weathered mafic rock (Figure 12 b; see Appendix 3);
- (iv) Very high concentrations of Ca, S and Ba occur in the weathered mafic rock, probably associated with cements and thick coatings of carbonate and barite (see Section 5.6.3).

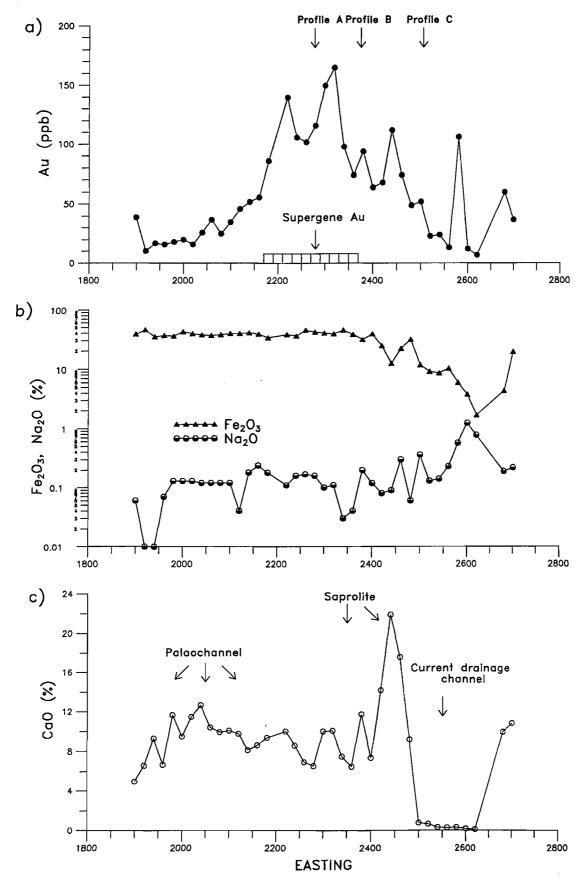


Figure 12: Distributions of a)Au, b)Fe $_2$ O $_3$ and Na $_2$ O, and c)CaO as shown by auger sampling 3700N.

5.4. Trench samples.

Trench samples from 3700N were collected and analysed for Au and As by Pancontinental Mining Ltd. Samples were taken at a constant level above the floor of the trench; the sampling depths were thus about 1.5m in the transported gravels and progressively shallower (to a few centimetres) as the trench crossed weathered rock.

Gold concentrations are highest in soils associated with the weathered rock whereas As concentrations are highest in the lateritic gravels (Figure 13). The Au anomaly in the trench samples is displaced with respect to those collected at the surface or by auger, and to the underlying supergene mineralization. This is because the carbonate horizon, in which the Au anomaly is located, was only sampled where the trench was shallow over the weathered rock (see Profile B) whereas in the transported gravels the barren horizon below the carbonate was sampled (see Profile A).

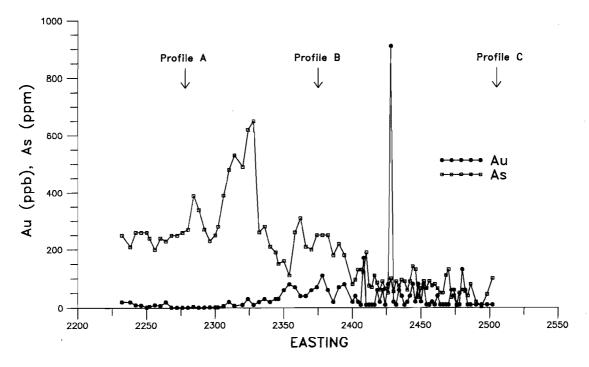


Figure 13: Distributions of Au and As in the trench on 3700N, as shown by sampling at a fixed height above the base.

5.5. Vegetation samples.

The occurrence of different plant species at Panglo is determined by soil type, geomorphology and lithology and herein lies the major problem with biogeochemical sampling in general. To make adequate comparisons between metal contents in plants along an entire traverse plants of the same species, age and condition should be sampled which is usually not possible in surveys of this type in Western Australia. However, this limitation was partly overcome by sampling where possible:

- (i) several species and grouping them according to their genera;
- (ii) Eucalyptus spp. and Eremophila spp. of similar trunk diameter and/or height.

Interpretation of the data are hindered because the soils were sampled by different procedures in different areas. Comparisons between the geochemistry of soils and vegetation should be made with caution since the horizon sampled may not reflect the zone used by the plants. This is particularly true for 4200N where samples were collected near the surface (5-20cm), which has considerable amounts of sand, and is unlikely to reflect the major zone of absorption of nutrients by the plants. Also, for 4200N, multi-element analyses of soils were restricted to a small part of the traverse (except for Au) limiting comparisons with plants.

Geochemical comparisons can be made more effectively between plant samples and soil samples on 3700N as augering of the top metre was performed. The root system of many shrub species e.g. Eremophila spp. occurs extensively within the top metre (rather than near the surface) and, consequently, they rely on penetration of surface rainfall for their water. However, Eucalyptus spp. can have, in addition to the near-surface root system, a long tap root (sometimes tens of metres in length) capable of absorbing nutrients and water from saprolitic clays and the water table, respectively i.e. they can be phreatophytic. Chemical contents of Eucalyptus spp., therefore, might represent a composite of that obtained by roots from two or more locations within the profile.

Neutron activation was initially used for the analysis of the plants, on a dry weight basis. However, the high Na contents of the plants caused a severe interference with Au, resulting in unacceptably high detection limits. The problem was circumvented by ashing the samples, digesting them in acid and analysing for Au by graphite furnace AAS.

Background Au concentration values for vegetation collected at about 4900N 2300E (see Figure 2), away from mineralization, averaged 0.036 ppm for Eremophila spp. and 0.012 ppm for Eucalyptus spp (see Appendix Section 5). Gold contents in vegetation at Panglo were generally lower than those found in other Western Australian studies (Smith and Keele, 1984; Lintern, 1989).

5.5.1. Traverse 4200N samples.

Eremophila spp. and Atriplex spp. occur most frequently on 4200N and, therefore, were sampled most frequently. Plant communities growing in soils developed over ultramafic/mafic rocks are open shrublands with Eremophila spp. and some Atriplex spp. whereas the communities growing over the sedimentary rocks which are mainly chenopods (saltbush). Geochemical differences are not great between individuals of the same genus growing on ultramafic rocks (generally located more to the western part of the traverse) and those growing on mafic rocks (Appendix 5).

Gold concentrations in vegetation are generally close to background and show at best a poor association with either cyanide-extractable Au from soils, lithology or economic mineralization (Figure 14a). Higher Au contents were found in *Eremophila* spp. (up to 0.024 ppm) and *Melaleuca* spp. (up to 0.132 ppm) as compared with *Atriplex* spp. (up to 0.018 ppm). The highest Au content (0.185 ppm) was recorded for *Dodonaea* sp. (a small halophytic shrub) but because its distribution was limited, only one specimen of this plant was sampled (Figure 14a) and so it is not known, therefore, whether this value is anomalous.

There are no significant differences in the compositions of Eremophila spp. and Atriplex spp., except for Ca, Na and Cu (Appendix 5), although Eremophila spp. are generally richer in all elements compared with Atriplex spp.

Calcium is an important macro-nutrient primarily required for mechanical strength. It is more richly concentrated in *Eremophila* spp. when compared with *Atriplex* spp. possibly because they require greater mechanical strength due to their height. Calcium concentrations are higher in the ash of *Eremophila* spp. than the soil. This may be of major importance when considering the origin of pedogenic carbonate.

Sodium is commonly present in large concentrations in plants adapted to growing in saline soils (e.g. the halophytes) and is involved in osmoregulation. Sodium concentrations are higher in Atriplex spp. than those in Eremophila spp.

Copper is a minor but essential plant nutrient contained in many enzymes and also in photosynthetic pathways. Copper is more richly concentrated in the ash of *Eremophila* spp. when compared with *Atriplex* spp. or the soil. The significance of this enrichment is not known.

5.5.2. Traverse 3700N samples.

Eremophila spp. and Eucalyptus spp. grow on the gravelly, well drained soils of the palaeochannel and those developed over mafic rocks. Halophytes grow in the clay-rich, poorly drained soils developed over the current drainage channel overlying sedimentary rocks (Figure 14b). Eremophila spp. and Eucalyptus spp. occur regularly along 3700N and were sampled most frequently (Appendix 5).

The distributions of Au shown by vegetation sampling are broadly similar to those found in soils; contrasts tend to be higher in gravelly soils. Gold contents are much higher in *Eremophila* spp. (up to 0.059 ppm) than *Eucalyptus* spp. (up to 0.042 ppm). *Eucalyptus* spp. east of the drainage channel contain more Au than *Eucalyptus* spp. found over mineralization to the west (Figure 14b) and may be of significance since they are also nearly double the background Au values for *Eucalyptus* spp. (0.012 ppm). Furthermore, the background data suggest that *Eucalyptus* spp. are not as effective as *Eremophila* spp. in accumulating Au. The Au content of *Eremophila* spp. on 3700N are generally richer than those on 4200N.

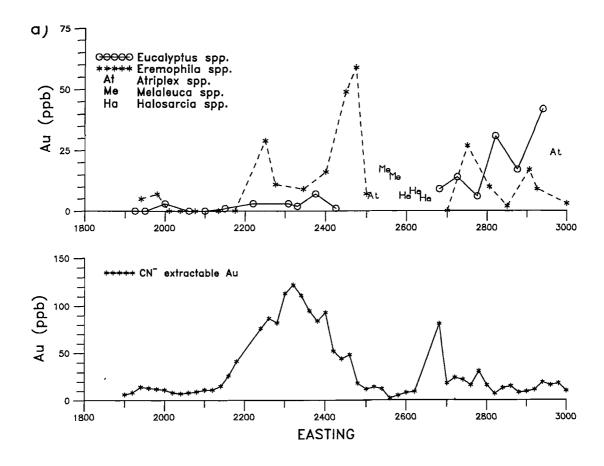
Apart from Au, there is little relationship between the composition of the vegetation and soils. This feature is most clearly seen with Fe, Mn, Ca, Ni and V: soil auger concentrations of these elements tend to decrease towards the drainage whereas in vegetation they tend to increase (Section 5.3; Appendix 3). Soils nearer the channel are thin because weathered rock occurs close to the surface. For Eremophila spp., the trend of slightly increasing concentrations of As, V and Ni, eastward and downslope towards the channel, may reflect the presence of these metals in forms more available for absorption by plants (e.g. from the breakdown of saprolite), in contrast to a less soluble form present in the gravelly soils. Sodium and ash concentrations are richest for plants in the channel vegetation samples and are comparable with halophytes found on 4200N.

The Cu contents of *Eremophila* spp. are higher than those in *Eucalyptus* spp. The values are comparable with those found for *Eremophila* spp. at 4200N and background areas, suggesting that appreciable quantities are required by this genus for normal plant function. Soil Cu contents are similar for the two traverses.

Sodium contents of *Eucalyptus* spp. are much greater than those found in *Eremophila* spp. but not as great as the salt-tolerant channel genera such as *Atriplex* spp. Sodium enrichment in *Eucalyptus* spp. may reflect the ability of its roots to reach, absorb and accumulate salts from the groundwater.

Calcium enrichment in *Eremophila* spp. is greater on 3700N than 4200N and may reflect higher abundance of pedogenic carbonate in the soil (Appendix 5).

Additional results for vegetation samples collected to the north of 3700N, but still in the gravelly soils, are recorded in Appendix 5. Metal concentrations are similar in magnitude to those found at locations on 3700N.



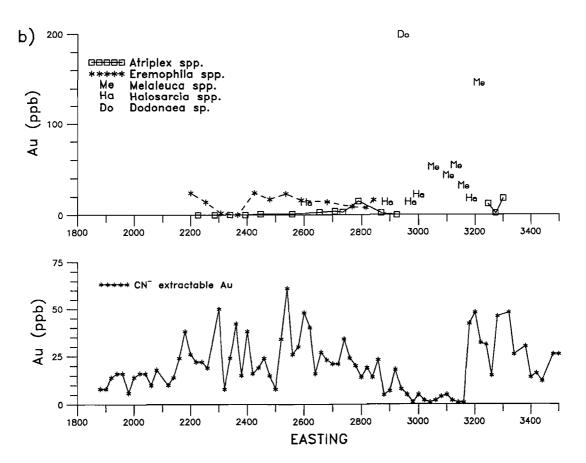


Figure 14: Au contents for vegetation and soils along traverse a) 3700N and b) 4200N

5.6. Special samples.

A more detailed examination of material similar to that found in Profile B was undertaken. It was important to establish the nature of the relationship between Au and carbonate in nodular material as the channel sampling procedure used in Profile B did not make the association clear. At the sample sites, the amount of carbonate coating the nodules varied in extent and thickness but tended to be greater closer to the soil surface. Lower in the profile the amount of soil and carbonate was poor or absent and the nodular material became more massive and indurated.

Special sample G1 (04-2112) was taken close to Profile B at 2377N 3700N from 30cm depth (Figure 2; Figure 3e). It comprises several sub-rounded and angular nodules up to 10 cm in diameter. Each nodule consists of a core of red ferruginized saprolite partially coated with pink coloured carbonate up to 3mm in thickness; some carbonate had penetrated into fine cracks in the nodules (Figure 15).

special sample G2 (04-2129) was taken from 2455E 3700N at 30cm depth, 80m east and downslope of the first. The profile was similar in appearance to that described for the first special sample but contained boulder-sized material. It consists of sub-rounded, dark red cobbles and boulders of ferruginous saprolite up to 40cm in diameter (Figure 3e). The red core is partly surrounded by a very pale layer varying from a few millimetres to several centimetres thick. The core and pale layer are referred to herein as "red saprolite" and "white saprolite" respectively. Veneers of pink-coloured carbonate covered areas of the pale material (Figure 16) but the total amount of carbonate is less than in sample 04-2112.

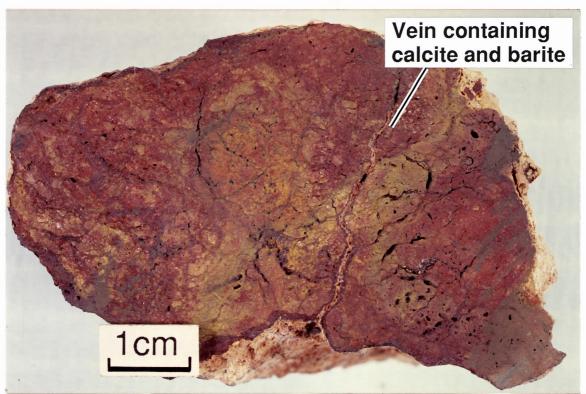


Figure 15: Sample G1 (04-2112). Photograph of a ferruginous carbonate-coated nodule. Note the prominent vein through the centre of the specimen which contains calcite and barite (see Figure 17).

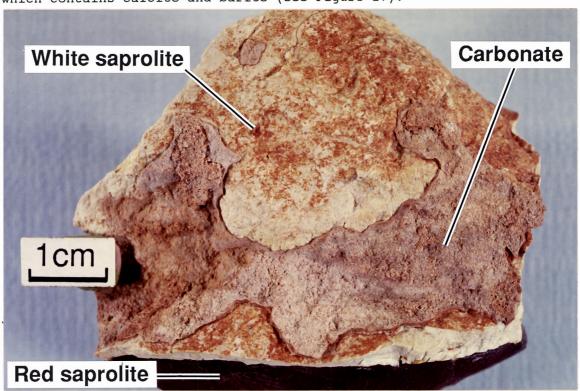


Figure 16: Sample G2 (04-2129). Photograph of a ferruginous nodule ("red saprolite") covered with pale saprolite ("white saprolite") and carbonate.

Representative photomicrographs, with interpretative diagrams, are shown in Figures 17 and 18. No visible Au was detected by the SEM which suggests that it is present as sub-micron particles.

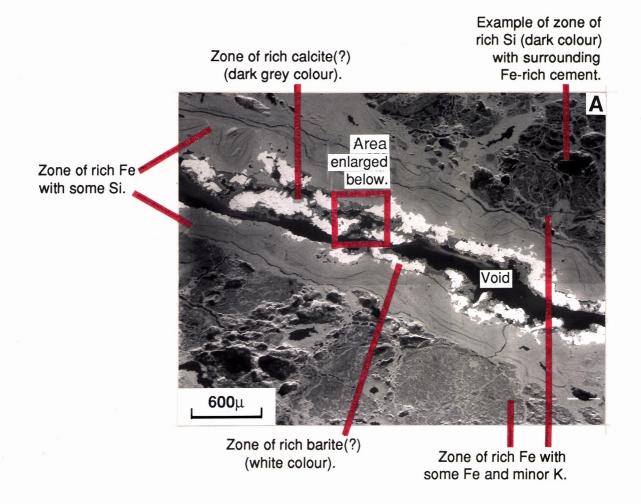
Geochemical studies were undertaken on sample G2 only, as carbonate skins on sample G1 were tooo thin to be removed from the nodule surfaces without contamination from underlying material. Sample G2 was divided into carbonate (pink), white saprolite and red saprolite using a small hand drill. Detailed composition of each type can be found in Appendix 6. The results are summarized and contrasted in the table below by comparing elemental concentrations found in the carbonate with the saprolites (Table 2).

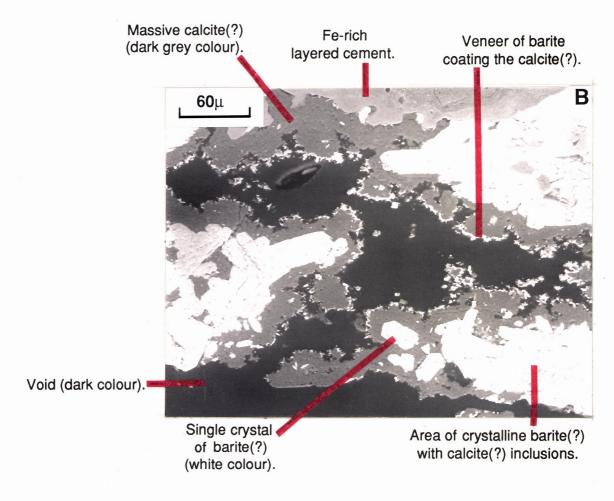
Amount	Carbonate	Red saprolite	White saprolite
High	Mg, Ca, Au, Ba,	Si, Al, Fe, Ni,	Si, Al, Ca, Na, Ti,
	Cu, Mn	Ti, Cr, V, Zr	Zr
Low	Si, Al, Fe, Na,	Mg, Ca, Au, Ba,	Fe, Mg, Au, Ba, Cr,
	Ti, Cr, V, Zr	Cu, Mn	Mn, V

Table 2: Comparison of element concentrations between the three types of material types.

Figure 17: Sample G1 (04-2122). Photomicrograph of a vein within a carbonate coated ferruginous saprolite nodule. The lower photomicrograph show details of the vein. Mineralogy estimated from SEM-EDAX analysis.

Figure 18: Sample G2 (04-2129). Photomicrograph of surface features of ferruginous saprolite nodules. The lower photomicrograph (c) show details of the surface for (b). Mineralogy estimated from SEM-EDAX analysis.





Calcite(?)-rich outer skin containing inclusions of quartz, Fe-rich concretionary material. — Also Ca-rich fragments containing Mg, Si and Fe.

Intermediate layer rich in Si with minor Al and Ca. Ca increasing towards outer calcite layer.
Barite(?) found near border of inner layer below.

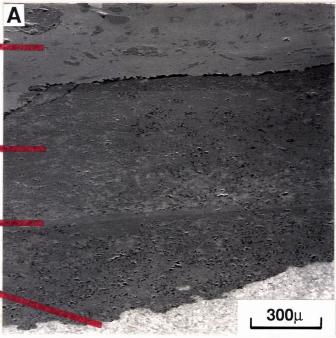
Si-rich structure within intermediate layer.

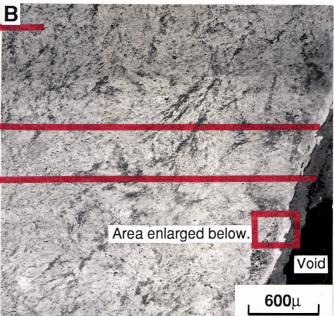
Fe-rich with some Si and trace
Ti and Al. Si and Al skins around
Fe-rich fragments found in detailed
viewing. Barite(?) found near
border of intermediate layer above.

Fe-rich saprolite with some Si and Al and trace Ca and K. Contains vein-like structures.

Fe-rich with some Ca, Si and Al.

Calcite(?) layer. Ba occasionally found at contact between calcite(?) and saprolite and also in veinlets extending into saprolite.



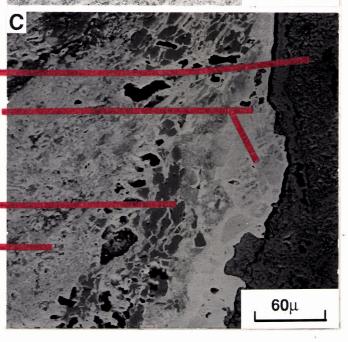


Cacite(?)-rich with trace Si and Fe. Ba found at contact between calcite(?) and saprolite and also in veinlets extending into saprolite.

Fe-rich becoming increasingly Ca-rich towards the calcite(?) layer and Si-rich towards the saprolite with some Al. Fe-rich pseudomorphs of Si-rich saprolite.

Si-rich saprolite with minor Fe and Ca.

Fe-rich with some Si and Al, and trace Ca and Ba.



There is a strong association between Au and the alkaline earth metals in these samples. The geochemical (Table 2) and photomicrograph (Figure 17) evidence suggests that Au has been transported in solutions containing Ca, Ba (and ${\rm SO_4}^{2-}$) and Mn. Low concentrations of these elements in the saprolite suggest that they are derived from elsewhere in the regolith. Laboratory studies suggest that Au is highly mobile in carbonate-rich soils (Gray et al., 1990).

Petrological and SEM studies of the contents of veins within sample G1 reveal their depositional sequence to be Fe (as an oxide), Ba (as barite) and Ca (as a low Mg calcite) (Figure 17). Barite occurrences tend to be greater deeper into the veins suggesting that early solutions were rich in Ba and sulphate. Later, pore waters rich in Ca were prominent. the third deposition stage, however, conditions once again favoured the formation of barite as it is found as a veneer on some of the calcite In general, however, calcite is much more abundant than (Figure 17b). Barium tends to be richer in the shales rather than the mafic and ultramafic rocks (Scott, 1990; Scott and Dotter, 1990). Sections of sample G2 reveal a similar paragenetic sequence of Fe-and Ca-containing solutions invading the host saprolite (Figure 18a and b). Iron-rich solutions have penetrated pores and, seemingly, replaced (pseudomorphed) some of the Si minerals near the edge of, and within, the saprolite (Figure 18c). Subsequently, solutions rich in Ca have followed and formed often thick veneers, covering saprolite surfaces and, occasionally, have entered weakened areas to form calcite-rich veins.

The presence of carbonate-rich skins and veins suggests a weathering process having both physical and chemical components. It probably involves (i) invasion by carbonate and replacement of host minerals, (ii) slow, and possibly repeated, crystallization and dissolution of calcite and, finally, (iii) fracturing by crystal growth. The rate of fragmentation will depend locally on many factors including the micro water-regime, soil temperature and the concentration of Ca and bicarbonate ion in the pore water. Chemical replacement by calcite has been reported for quartz (Pettijohn, 1972; Millot et al, 1977) and Fe oxides (Nahon et al, 1977).

6. Discussions and Conclusions.

6.1. The Model.

The results demonstrate a strong association between Au and pedogenic carbonate. The association is particularly strong when the concentrations of Ca are greater than one or two percent. This result is consistent with that found in similar studies conducted at Mt. Hope (the Bounty mine), W.A. (Lintern, 1989; Lintern et al, 1990). The particular significance of the present study is the presence of Au-bearing, near-surface carbonate in 10m of transported overburden, which is, in turn, 30m above the main zone of supergene mineralization (Profile A; Figure 19). It is presumed that the Au is vertically derived from this mineralization either directly, or interface between the indirectly via the enrichment zone at the palaeochannel and the post-lateritic land surface at 10m depth (Figure 19). Similar Au anomalies developing in 4-6m of transported overburden have been reported by Lawrance (1988).

The precise mechanism by which the Au and calcite are associated has not However, the association between Au concentrations in been determined. vegetation and those in soils, and the high concentrations of Ca found in both vegetation and soil, suggests Au and Ca are absorbed by deep-rooted trees and plants, translocated to the leaves and deposited in soils by leaching or leaf fall. Subsequently, multiple recycling of the Au and Ca within the soil continues, by the same mechanisms, with less deeply rooted The initial co-location of Au and Ca is thus due to their uptake and association within the vegetation. The model may also explain why higher Au contents are generally found in soils beneath woodland rather than soils beneath shrubland since the trees are deep rooted and have a greater opportunity to continue to supplement surface soils with Au derived Laboratory experiments at CSIRO support the model, as they from depth. suggest that Au is relatively soluble and mobile in soil environments (Gray et al., 1990). The accumulation of carbonate in soils occurs in semi-arid climates where evaporation and evapo-transpiration exceed precipitation. Partially soluble elements such as Ca remain close to their source i.e. from Ca-containing rocks or groundwaters. At Panglo, the nearest present source of Ca are the groundwaters (Gray, 1990) which are at 5 to 10m depth.

The chemistry of the association between carbonate and gold may involve a complex series of steps (see Milnes and Hutton, 1983). One possible mechanism is as follows:

- (i) Au and Ca, derived from vegetation and the former complexed as an organic ligand, are dissolved by meteoric water. The solubility of Ca^{2+} depends on the partial pressure of CO_2 and temperature.
- (ii) High concentrations of ${\rm CO}_2$ occurring in the soil (derived from soil micro-organism and root respiration) are dissolved in the water.

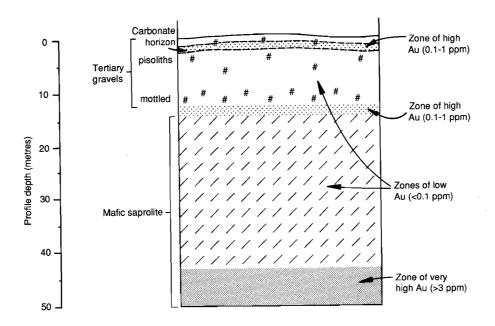


Figure 19: Stylized section through 3700N showing relative positions of Au mineralization in (i) the carbonate horizon, (ii) the post-lateritic land surface and (iii) the economic mineralization at 40m.

- (iii) Water is removed from the soil by evaporation and roots, as it percolates downwards.
- (iv) Ca²⁺ concentrations reach saturation and precipitate. Calcium precipitates as carbonate and Au becomes adsorbed onto it as an organic complex. The chemistry of precipitation of organic-Au complexes under these conditions are poorly understood.

(v) With the introduction of further water the dissolution process begins again. Laboratory experiments show that appreciable soil Ca and Au are soluble under normal atmospheric conditions (Gray et al., 1990) and for Ca, at least, solubilities are increased with increased $\rm CO_2$.

The downwards movement of Ca and Au in the profile to their present locations, therefore, is a result of an iterative process dependant on the percolation of water from a series of rainfall events and recycling via plant roots.

Carbonates are generally only retained in the landscape in regions in which rainfall is less than 600mm (Young, 1973). Pedogenic carbonates tend only to be abundant in areas of winter rainfall in which there is a long growing season and an extended period of high CO₂ (Butt et al., 1978). rainfall areas, carbonates tend to form valley calcretes. It is not clear why carbonates may occur as powdery nodules, massive boulders or only as thin coatings but probably involves a combination of soil processes and include: porosity, mineralogy, vegetation, These Carbonate-enriched soil horizons groundwater, temperature and rainfall. may be entirely absent, or separated from Au enrichment for a variety of reasons. These may include, for example, different (i) soil moisture, (ii) soil types, (iii) soil temperature, (iv) biological activity, (v) climatic regime and (vi) lithology. More data are obviously required on this topic. Although above background Au values are generally found in calcite, Au maxima concentrations in some soil profiles (e.g. Profile C) do not strictly correspond with Ca maxima concentrations. Carbonate-poor soils are also extremely common throughout Western Australia. In wet areas, for example those occurring at the Boddington gold mine in the Darling Range of Western Australia, Ca concentrations in laterite soils are poor whilst Au concentrations may still be rich. In desert areas, low ${\rm CO}_2$ activity in the soil may lead to the formation of aluminosilicates or gypsum in place of Clearly, further work is required to establish how calcrete forms and to investigate the mineralogical and geochemical associations of Au in soils in other environments.

6.2. Implications for exploration.

6.2.1. Soil sampling procedures.

The strong association of Au with pedogenic carbonate implies important exploration sample extremely are an Carbonates hold information on underlying Au mineralization even in areas of transported overburden and sampling provides, therefore, a low cost alternative to other techniques such as scout RAB drilling to bedrock. is not certain that the association is present in all calcareous horizons. However, if present, it is important that the carbonate horizon is identified and correctly sampled. The study has shown that the depth of the main carbonate horizon can vary. Simple field test procedures using dilute hydrochloric acid can unequivocally identify the presence of soil Dolomite is less reactive than calcite and takes slightly carbonate. longer to effervesce. Field technicians should be encouraged to use the test and document the presence or absence of carbonate as part of routine soil sampling surveys, to assist in their interpretation.

Auger drilling and compositing the top metre will effectively sample the carbonate horizon in most areas, although some dilution from carbonate-poor horizons is to be expected. The size and signal response can be increased by coring the soil and selecting the richest carbonate zone, but this is probably operationally impractical.

In surface soil sampling particular attention should be paid to the presence of carbonate material at or near the surface as concretionary nodules or coatings. Commonly, carbonate is found directly on the surface and will be effectively sampled by this procedure (as in 4200N) but it also may be covered by shallow, recently transported material. Differences in sample depth of just a few centimetres may thus change the response for Au. If the carbonate is present as coatings on rocks or gravels, then the volume of the coated material should be considered. Large sized fractions (e.g. gravels) have a small surface area to volume ratio and will consequently give a low response.

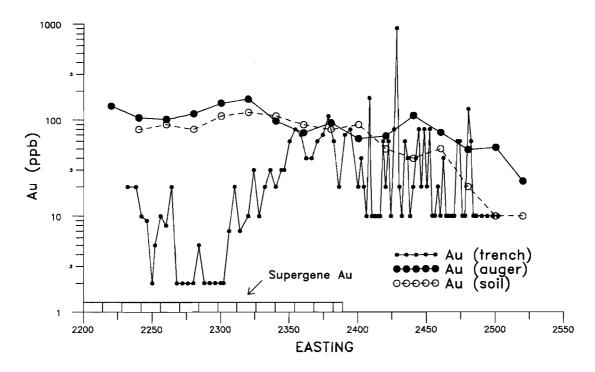


Figure 20: Comparison of Au results using different soil sampling procedures.

Of all the soil sampling procedures examined, trench sampling appears to be the most vulnerable to mis-interpretation yet potentially offers the greatest amount of information (Figure 20). The results have demonstrated how the depth of the trench is important for sampling. Trench sampling is performed by taking channel samples at a standard height (usually knee or waist) from the bottom of the trench. As carbonate is usually found within a metre of the surface, a 2m deep trench will be incorrectly sampled by the "standard" procedure and may miss the Au anomaly altogether. In a 1m (or less) deep trench samples are likely to contain carbonate and will be correctly sampled. The trench at 3700N becomes shallower from the gravels (e.g. at Profile A) to the partially weathered rock (e.g. at Profile B) as the earthmoving equipment cannot penetrate the harder material. "standard" procedures only samples the carbonates in the shallow part of trench and so incorrectly locate the position of mineralization (assuming vertical migration of Au). The error of location at 3700N is a few tens of metres but in certain circumstances a trench anomaly may be ignored as being false and underlying mineralization missed completely.

6.2.2. Vegetation sampling.

The importance of the role of vegetation sampling in gold exploration at Panglo appears less well defined and are attributable to biogeochemical sampling problems in general. The composition of vegetation reflects, to a great extent, a) the availability of the element in the vicinity of the root system and b) the ability of the plant to accumulate elements within its cell structures. Other more specific considerations for biogeochemical sampling include the particular plant organ sampled e.g. leaves, roots or branches, the age and health of the plant, genetically-determined individual variation and the time of year sampled. The major perceived advantage that biogeochemistry has over soil sampling is that it can, indirectly, potentially sample greater depths.

Despite the presence of Au in vegetation, it has been shown that biogeochemistry alone cannot be used as an exploration tool at Panglo as it does not correctly define the location of underlying mineralization on either traverse. There is some evidence that on 3700N, there is a relation between Au in vegetation and Au in soils, but on 4200N, there is no indication that the Au content of vegetation reflects either Au in soils or mineralization at depth. Vegetation Au contents are generally lower at Panglo relative to those reported at Mt. Hope (Lintern, 1989) and Norseman (Smith and Keele, 1984) particularly over mineralization. This probably reflects the depth to mineralization which at Panglo is about 40m and at Mt. Hope is within a few metres of the surface. If vegetation is to be sampled it was found at both Panglo and Mt. Hope that the shrub species (Eremophila and Melaleuca spp.) have higher Au concentrations than the trees (Eucalyptus spp.) over mineralization.

Carbonate-enriched soil horizons may be entirely absent, or separated from Au enrichment for a variety of reasons. These may include, for example, different (i) soil moisture, (ii) soil types, (iii) soil temperature, (iv) biological activity, (v) climatic regime and (vi) lithology. More data are obviously required on this topic. Although above background Au values are generally found in calcite, Au maxima concentrations in some soil profiles (e.g. Profile C) do not strictly correspond with Ca maxima concentrations.

6.2.3. Analytical methods.

The results of the cyanide extraction of unpulverized samples indicate that a large proportion of the Au found in soils is easily dissolved. not surprising as laboratory studies have shown that some soil Au will dissolve in water (Gray and Lintern, 1990). The cyanide extraction method has several advantages as a routine analytical procedure for exploration over other methods currently used such as fire assay fusion and aqua regia digestion (Fletcher and Horsky, 1988; Lintern, 1988). These advantages include opportunities to (i) reduce contamination risk from laboratory sample preparation procedures e.g. sample crushing and pulverization, (ii) take a larger more-representative sample size, (iii) to perform in-field preparation (and analysis, see Lintern, 1988) of soil digest solutions e.g. a few millilitres ready for analysis thus circumventing the need to transport several kilos of solid samples (iv) potentially lower analytical The main disadvantage of cyanide digestion is the uncertainty of not knowing whether a large proportion of the Au has been solubilized This concern can be resolved, in part, by the inclusion of several check samples retained in solid form for total Au analysis.

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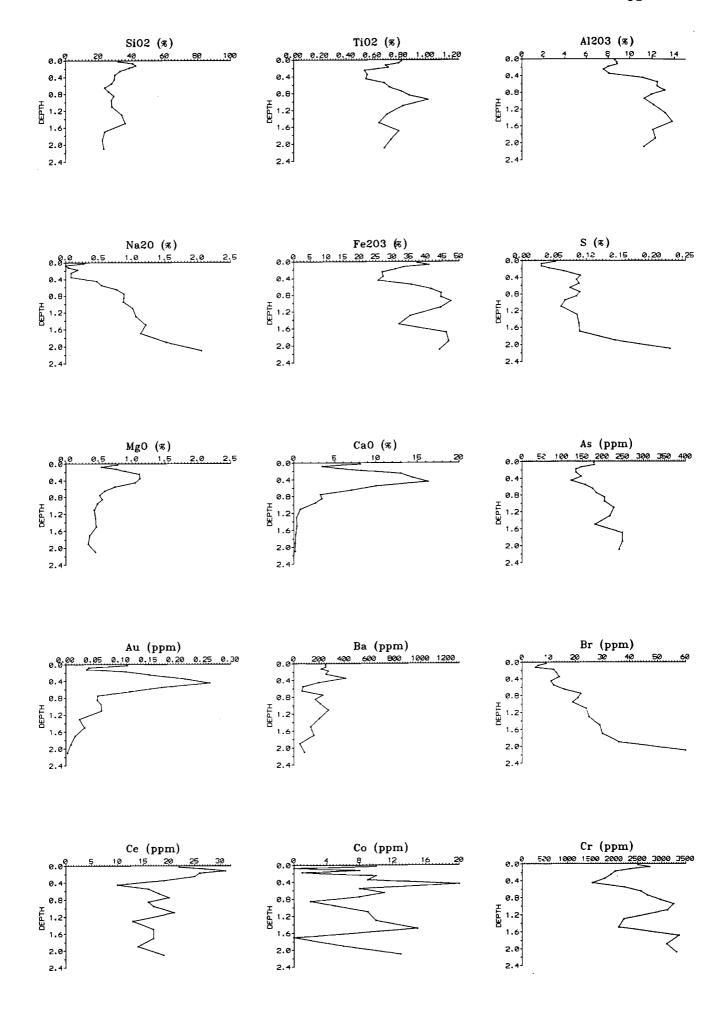
8. ACKNOWLEDGEMENTS.

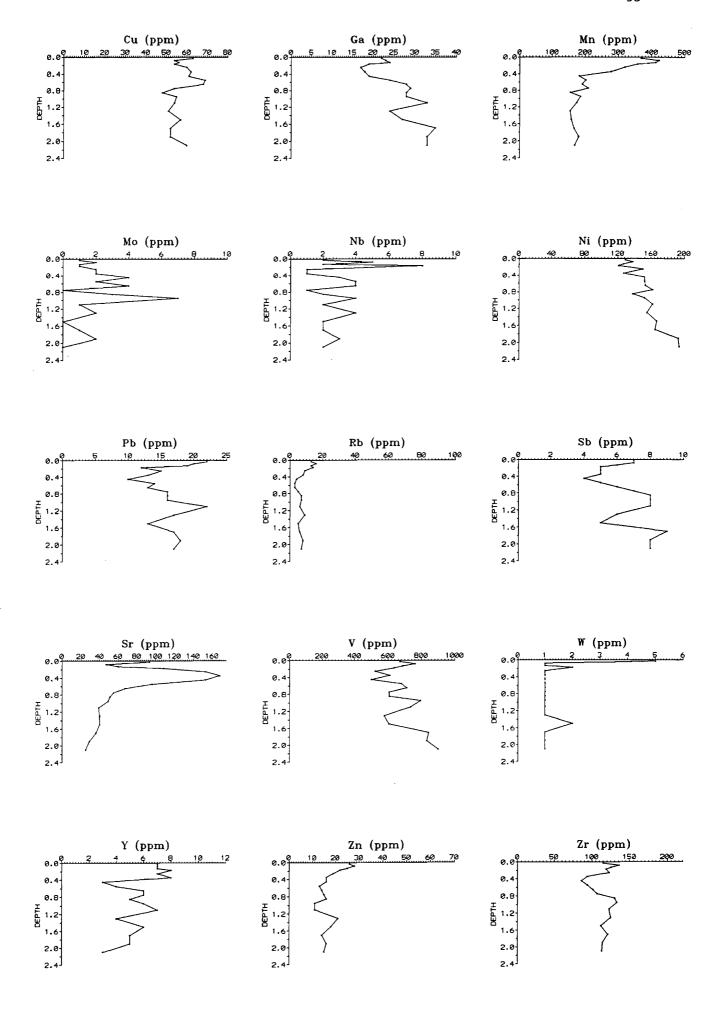
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- 9. APPENDICES.
- 1. Profile A-E samples: Tabulated and graphed data.
- 2. Near-surface samples: Tabulated and graphed data for 4200N.
- 3. Auger samples: Graphed data only.
- 4. Trench samples data: Tabulated data only
- 5. Vegetation sample data: Tabulated and graphed data.
- 6. Grab samples: Tabulated data.
- 7. Tabulated data for Profile C samples showing Au distribution analysis on two size fractions.

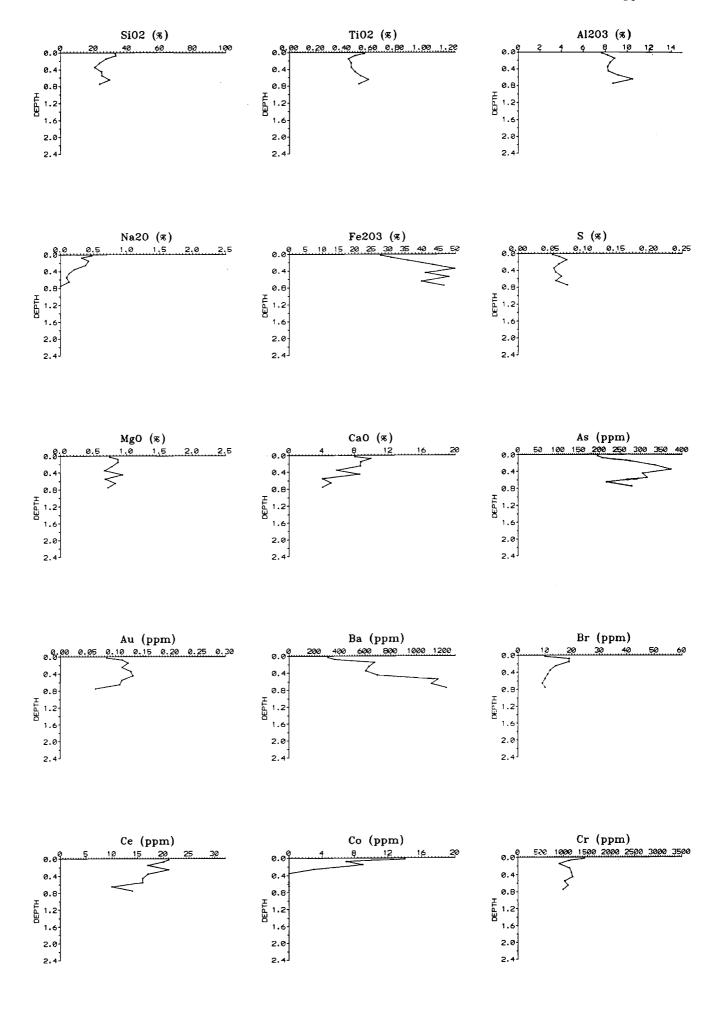
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SAMPLE 04-2085 04-2086 04-2087 04-2088 04-2090 04-2091 04-2092 04-2093 04-2095 04-2097 04-2098 04-2099 04-2100 04-2101	As 173 150 148 135 148 123 158 176 185 205 205 227 218 182 249 249 241	Au 0.111 0.042 0.038 0.109 0.152 0.217 0.262 0.174 0.116 0.058 0.057 0.064 0.065 0.025 0.034 0.017 0.009 0.003	Ba 224 225 204 241 294 426 234 74 90 253 175 205 348 196 137 131 43 70	Be 1 1 0 0 0 2 1 1 1 1 3 4 2 2 2 3 3	Br 9 6 5 12 13 14 11 12 16 22 21 19 24 25 29 30 36 60	Ce 22 27 31 26 25 19 10 16 18 20 16 17 21 13 17 17 14 19	Co 10 0 8 1 10 9 20 8 11 8 2 5 9 10 15 0 6	Cr 2501 2742 2337 2021 1931 1793 1537 2218 2568 2708 2969 3249 3119 2210 2098 3362 3106 3306	Cu 63 54 56 54 60 62 61 69 68 54 48 55 54 51 57 52 52 60	Ga 22 23 24 19 17 18 19 24 28 29 28 28 33 24 27 35 33 33	Ge 0 1 1 2 0 0 0 0 3 1 2 2 2 0 0 0 0 1 0 0	Mn 368 423 412 358 320 278 182 202 191 209 155 186 174 154 158 166 180 168
SAMPLE 04-2085 04-2086 04-2087 04-2088 04-2090 04-2091 04-2092 04-2093 04-2094 04-2095 04-2096 04-2097 04-2098 04-2099 04-2100 04-2101 04-2102	Mo 1 2 1 1 2 2 4 2 4 0 3 7 1 2 0 1 2 0	Nb 2 5 2 8 1 1 3 4 4 1 2 4 2 2 3 2 2	Ni 129 138 127 121 150 127 152 152 153 162 138 152 162 155 167 165 193	Pb 22 20 19 12 15 13 10 14 13 16 16 22 17 13 17 18 17	Rb 13 16 13 14 9 8 4 4 3 3 5 7 7 6 6 9 5 6 8 8 7	Sb 7 7 6 5 5 5 4 5 6 7 8 8 8 6 5 9 8 8 8	Sr 93 47 60 108 153 168 153 96 67 55 51 49 39 40 40 36 29 25	V 671 762 688 636 522 610 499 680 711 606 607 794 735 577 606 843 833 901	W 5 1 1 2 1 1 1 1 1 1 1 2 1 1 1 1 1 1 1 1	Y 7 7 7 8 7 8 3 4 6 6 6 5 6 6 7 4 6 6 5 5 5 3	Zn 26 28 25 22 19 16 16 13 14 15 16 11 11 21 18 14 16	Zr 115 136 125 119 123 94 86 94 101 107 130 133 123 125 112 121 114

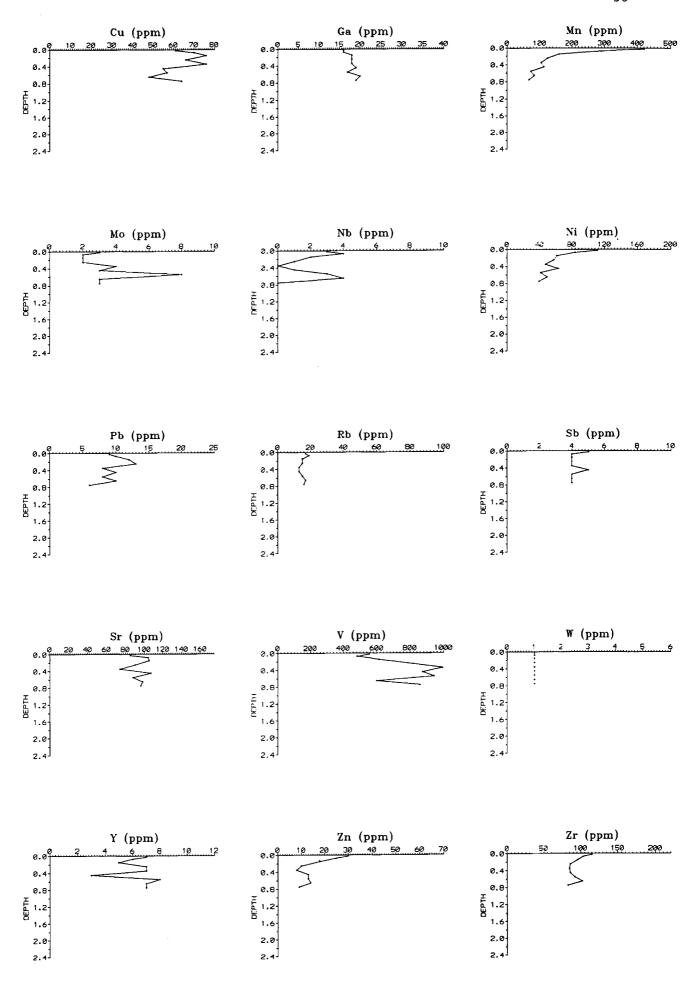
Appendix 1: Tabulated and graphed data for Profile A.





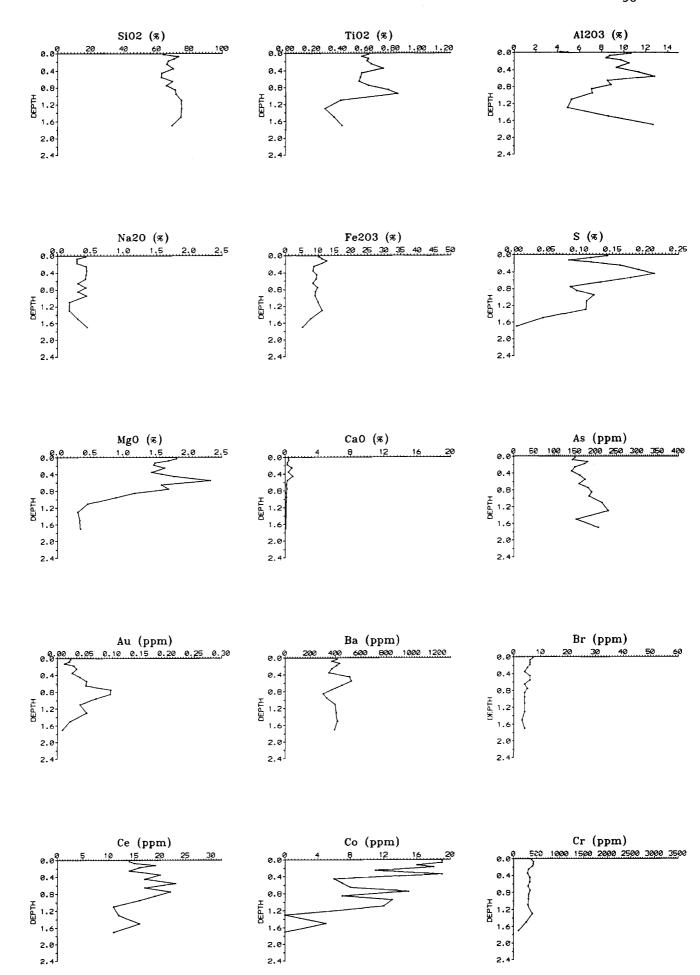
SAMPLE 04-2103 04-2104 04-2105 04-2106 04-2107 04-2108 04-2109 04-2111	2375 2375 2375 2375 2375 2375 2375 2375	NORTHING 3700 3700 3700 3700 3700 3700 3700 370	DEPTH(m) 0.025 0.075 0.150 0.250 0.350 0.450 0.550 0.650 0.750	\$i02 32.95 33.32 27.38 23.49 20.63 24.93 24.98 29.52 23.56	TiO2 0.57 0.50 0.45 0.47 0.47 0.50 0.54 0.60 0.53	A12O3 7.65 8.20 8.84 8.43 8.20 8.26 9.15 10.49 8.70	No20 0.47 0.32 0.42 0.38 0.21 0.13 0.09 0.13	Fe203 27.72 30.72 35.81 43.16 49.78 41.04 47.94 39.91 46.49	\$ 0.053 0.064 0.075 0.063 0.055 0.058 0.067 0.058 0.076	MgO 0.75 0.87 0.87 0.78 0.67 0.94 0.68 0.83 0.72	CaO 8.03 9.94 8.73 8.69 5.78 8.59 4.07 5.09 4.07	P205 0 0 0 0 0 0 0
SAMPLE 04-2103 04-2104 04-2105 04-2106 04-2107 04-2108 04-2110 04-2111	As 195 207 275 335 373 305 316 217 278	Au 0.084 0.112 0.123 0.112 0.128 0.132 0.112 0.108 0.064	Ba 252 334 770 658 702 825 1181 1192 1085	Be 2 1 3 2 2 0 0 0 0 2	Br 10 19 19 14 12 11 10 9	Ce 21 20 17 21 17 16 16 10 14	Co 14 7 9 3 0 0 0	Cr 1435 1115 901 1125 1163 1183 1018 1093 984	Cu 61 70 76 66 76 55 57 48 64	Ga 16 16 18 18 18 19 17 20	Ge 1 0 1 3 0 1 1 2	Mn 420 289 159 124 104 112 72 83 67
SAMPLE 04-2103 04-2104 04-2105 04-2106 04-2107 04-2108 04-2109 04-2110 04-2111	Mo 3 2 2 2 2 4 3 8 3 3	Nb 3 4 2 1 0 1 3 4 0 0	Ni 111 84 61 57 47 63 41 49	Pb 9 10 12 13 8 10 8 10 6	Rb 17 19 15 15 13 13 15 17 16	Sb 5 4 4 4 5 5 4 4 4	Sr 85 104 105 90 74 107 88 98 96	V 553 482 612 815 993 874 941 599 857	W 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Y 7 6 5 7 7 3 8 7	Zn 30 25 18 10 8 13 13 14	Zr 115 103 96 86 85 86 92 102 83

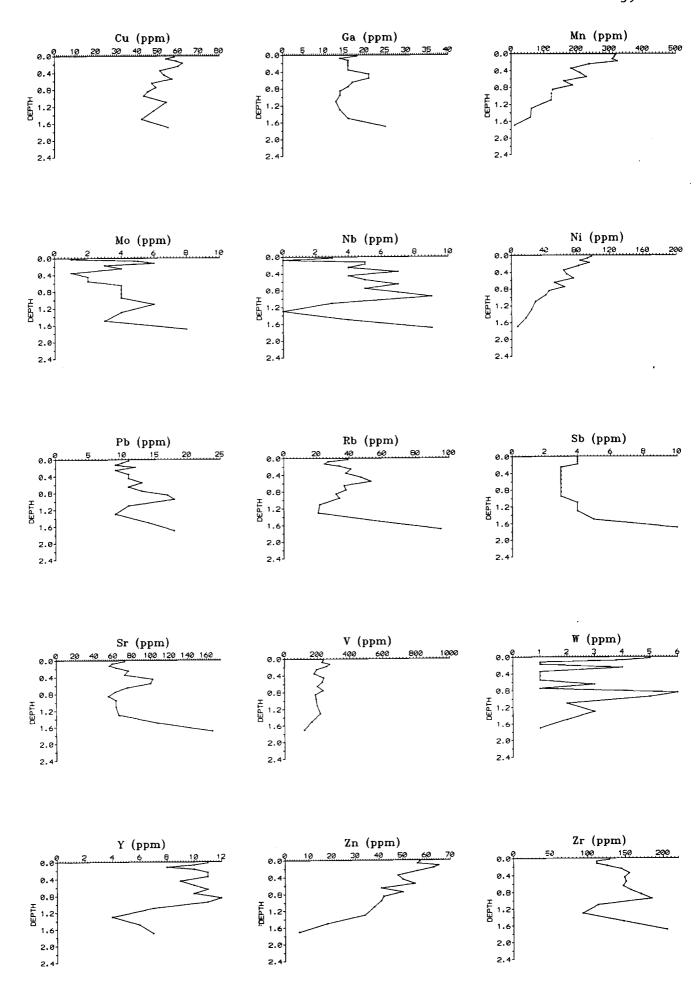




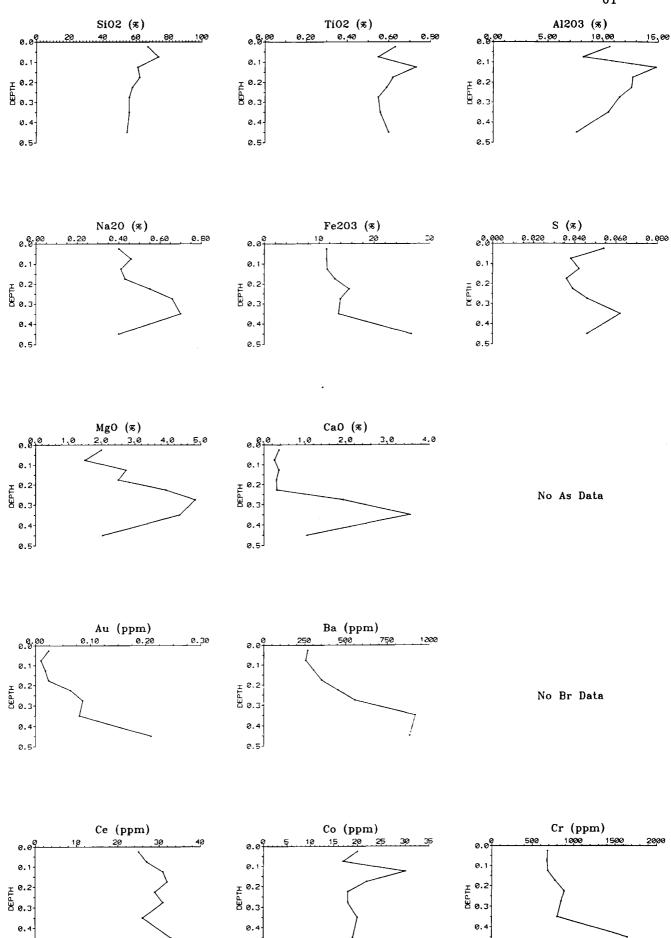
SAMPLE 04-2113 04-2114 04-2115 04-2116 04-2117 04-2118 04-2119 04-2120 04-2121 04-2122 04-2123 04-2124 04-2125 04-2125 04-2126 04-2127 04-2128	2505 2505 2505 2505 2505 2505 2505 2505	NORTHING 3700 3700 3700 3700 3700 3700 3700 370	DEPTH(m) 0.025 0.075 0.125 0.175 0.250 0.350 0.450 0.550 0.650 0.750 0.850 0.950 1.100 1.300 1.700	SiO2 64.05 73.28 70.63 67.11 66.20 70.06 63.02 62.83 69.46 65.92 71.27 71.69 75.09 75.19 74.75 69.27	TiO2 0.63 0.58 0.63 0.62 0.65 0.74 0.58 0.57 0.56 0.63 0.78 0.85 0.42 0.30 0.37 0.43	Al203 10.64 8.70 8.41 9.72 10.48 9.31 11.31 12.79 8.52 8.82 7.08 7.14 5.26 4.90 8.59 12.65	No2O 0.42 0.29 0.29 0.29 0.43 0.44 0.43 0.42 0.31 0.43 0.31 0.44 0.18 0.18 0.31 0.45	Fe2O3 10.19 11.13 12.61 11.31 8.67 8.37 9.57 9.36 8.38 9.87 9.21 9.16 10.04 11.26 7.72 5.30	\$ 0.142 0.117 0.084 0.118 0.161 0.189 0.213 0.177 0.131 0.086 0.096 0.122 0.111 0.11 0.045 0.004	MgO 1.81 1.69 1.48 1.46 1.63 1.77 2.33 1.58 1.69 1.17 0.89 0.46 0.31 0.34	CoO 0.42 0.44 0.29 0.26 0.82 0.39 0.94 0.28 0.18 0.21 0.16 0.15 0.09 0.08	P205 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
SAMPLE G=2113 G=2114 04-2115 04-2116 04-2117 04-2118 04-2119 04-2121 G=2122 04-2123 04-2124 G=2125 04-2126 G=2127 04-2128	As 150 144 180 173 150 142 162 174 160 182 190 184 215 230 153 206	Au 0.024 0.022 0.013 0.030 0.035 0.027 0.041 0.053 0.052 0.097 0.096 0.070 0.041 0.053 0.023 0.009	Ba 395 401 446 419 352 344 495 537 430 376 287 307 393 413 426 361	Be 1 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	Br 7 6 6 6 5 4 6 6 4 5 4 4 4 4 3 4	Ce 14 15 19 16 14 20 17 23 17 22 19 16 11 12	Co 19 19 16 18 11 19 6 7 8 15 7 13 12 0 5	Cr 385 429 424 413 322 299 349 350 310 357 322 321 308 408 275 105	Cu 58 54 59 62 60 51 53 57 47 49 45 43 54 48 42 55	Ga 18 14 16 16 16 16 21 21 17 16 14 14 13 14 16 25	Ge 1 1 2 2 2 1 1 2 2 1 1 2 2 3	Mn 318 314 309 324 239 184 211 230 162 188 128 123 122 61 59 10
SAMPLE G4-2113 G4-2114 G4-2115 C4-2116 C4-2117 G4-2119 G4-2120 C4-2121 C4-2122 C4-2122 C4-2123 G4-2124 G4-2125 G4-2125 G4-2127 G4-2127	Mo 1 5 6 3 4 1 2 2 4 4 4 4 6 6 4 3 8	Nb 3 0 5 5 4 7 4 5 7 9 3 0 4 9	Ni 98 93 84 95 82 64 68 76 53 65 46 42 29 24	Pb 11 10 9 12 9 11 11 13 11 13 17 18 11 9 14 18	Rb 39 27 25 34 41 38 47 53 37 38 32 34 22 21 58 95	Sb 4 4 4 4 3 3 3 3 3 3 4 4 4 5 10	Sr 72 59 56 63 76 72 102 100 75 62 55 63 66 107 166	v 236 229 276 258 195 181 239 231 200 236 188 192 199 219 166 120	W 5 4 1 1 4 1 1 1 3 1 6 5 2 3 2 1	Y 11 10 8 10 11 11 9 10 11 10 12 11 7 4 6 7	Zn 57 56 63 56 48 50 55 41 50 42 41 38 34 18 6	Zr 129 112 112 126 145 155 149 151 147 157 171 185 114 94 148 205

Appendix 1: Tabulated and graphed data for Profile C.





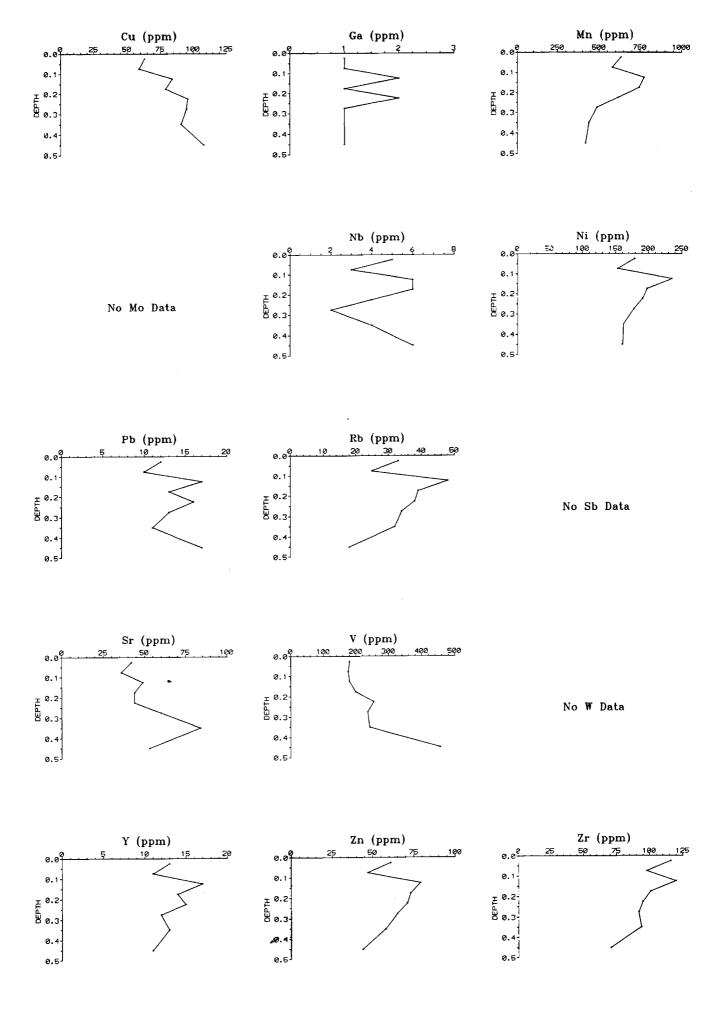
SAMPLE 04-2200 04-2201 04-2202 04-2203 04-2204 04-2205 04-2206 04-2207	EASTING 2420 2420 2420 2420 2420 2420 2420 242	NORTHING 4200 4200 4200 4200 4200 4200 4200 420	DEPTH(m) 0.025 0.075 0.125 0.175 0.225 0.275 0.275 0.350 0.450	SiO2 67.0 73.6 61.1 62.2 57.9 56.0 55.9 54.6	TiO2 0.63 0.55 0.73 0.62 0.59 0.55 0.56 0.60	Al2O3 10.6 8.2 14.8 12.7 12.6 11.5 10.5 7.6	No2O 0.22 0.22 0.28 0.27 0.35 0.36 0.37 0.16	Fe2O3 11.4 11.4 11.5 12.9 15.5 13.9 13.6 26.7	\$ 0.054 0.038 0.042 0.036 0.039 0.046 0.062 0.046	MgO 1.99 1.50 2.73 2.50 3.94 4.83 4.36 2.03	CaO 0.37 0.26 0.37 0.31 0.32 1.93 3.54 1.05	P205 0.08 0.05 0.09 0.05 0.04 0.05 0.04
SAMPLE 04-2200 04-2201 04-2202 04-2203 04-2204 .04-2205 04-2206 04-2207	As	Au 0.024 0.010 0.018 0.024 0.064 0.086 0.080 0.210	Ba 268 256 304 354 453 554 916 882	Be 0.0 0.0 0.4 0.0 0.0 0.0 0.0 0.0 0.0 0.0	Br 	Ce 25 27 31 32 29 31 26 33	Co 20 17 30 22 18 18 20	Cr 683 677 687 776 884 848 802 1650	Cu 63 59 84 79 96 95 91	Ga 1 1 2 1 2 1 1 1 1	Ge 0 0 2 0 2 1 0 0 0	Mn 679 634 829 785 643 515 465 426
SAMPLE 04-2200 04-2201 04-2202 04-2203 04-2204 04-2205 04-2206 04-2207	Mo - - - - -	Nb 5 3 6 6 4 2 4 6	Ni 166 151 243 213 206 183 173	Pb 12 10 17 13 16 13 11 17	Rb 33 25 48 39 38 34 32	Sb	Sr 42 36 49 44 44 60 84 53	v 210 201 209 228 285 274 285 547	W	Y 13 11 17 14 15 12 13	Zn 61 47 79 73 71 65 58	Zr 116 98 120 101 95 92 94 71



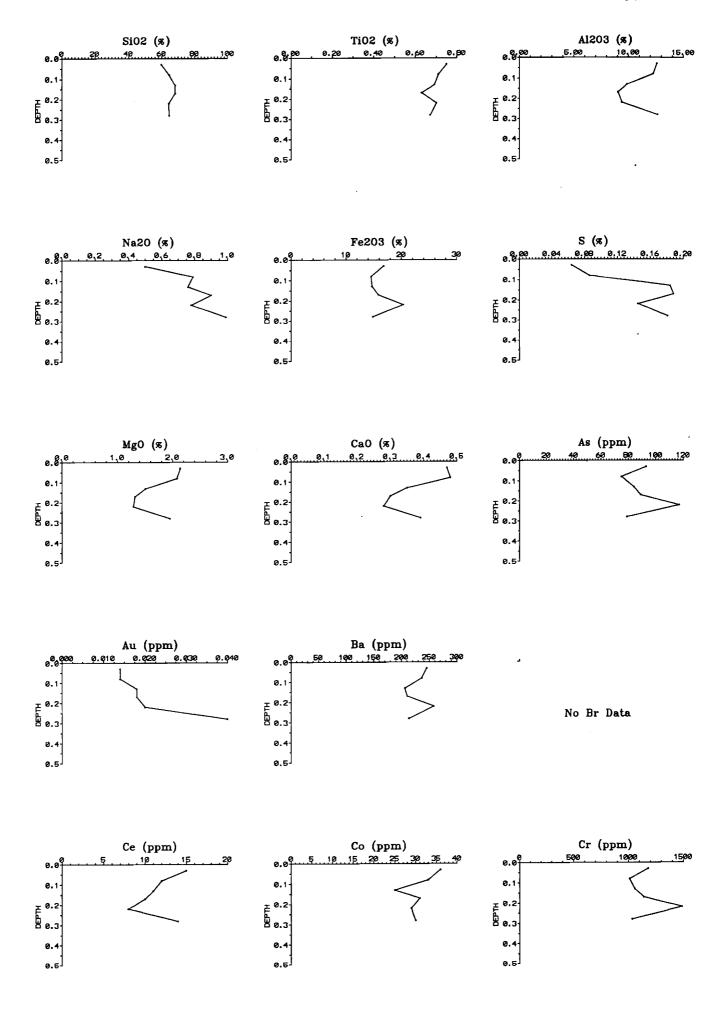
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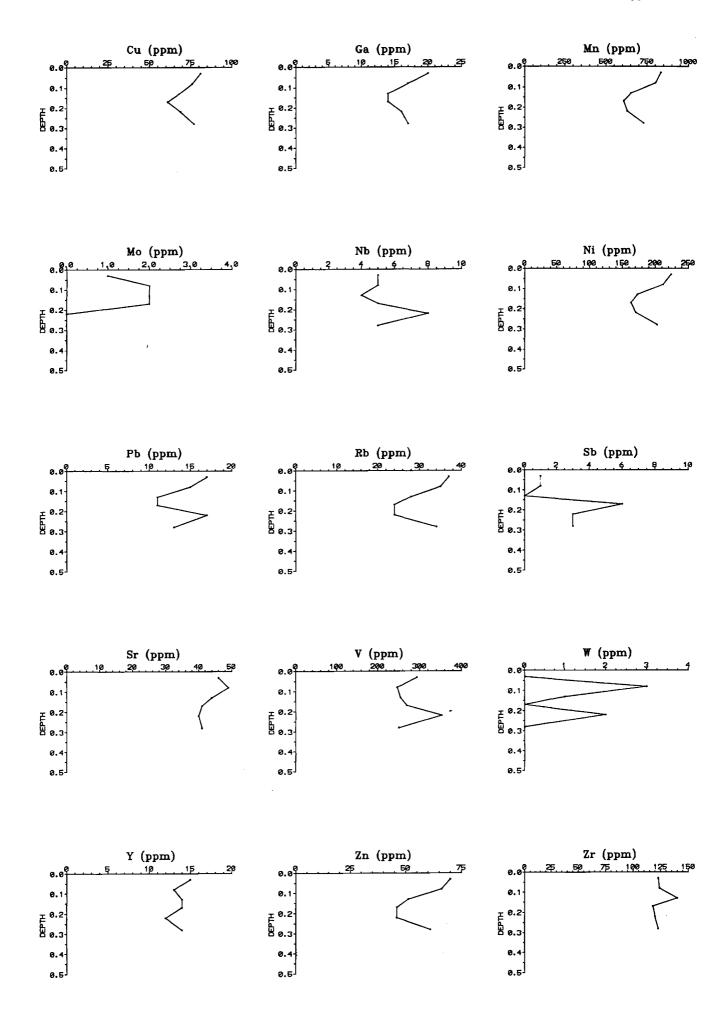
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0.5



SAMPLE 04-2217 04-2218 04-2219 04-2220 04-2221 04-2222	2340 2340 2340 2340 2340 2340 2340	NORTHING 4300 4300 4300 4300 4300 4300	DEPTH(m) 0.025 0.075 0.125 0.175 0.225 0.275	SiO2 59.8 64.57 68.28 68.28 64.34 64.57	7iO2 0.71 0.68 0.68 0.62 0.67 0.66	Al203 12.59 12.27 9.88 9.04 9.41 12.65	Na20 0.29 0.4 0.4 0.3 0.4 0.63	Fe2O3 16.78 14.55 14.67 15.82 20.29 14.84	\$ 0.064 0.086 0.184 0.188 0.145 0.181	MgO 2.14 2.08 1.51 1.32 1.29 1.95	CoO 0.47 0.48 0.35 0.3 0.28 0.39	P205 0.06 0.04 0.05 0.01 0.03 0.04		
SAMPLE 04-2217 04-2218 04-2219 04-2220 04-2221	Ag 2 0 1 1 1 4 1	As 93 75 84 89 117 79	Au 0.014 0.014 0.018 0.018 0.02 0.04	Ba 251 242 208 217 257 220	Be 0.559 0.86 0.009 0.005 0.259 0.301	Bi 2 2 0 0 3 1	Ce 15 12 11 10 8 14	Co 23 29 17 18 13	Cr 1085 944 993 1067 1337 972	Cs 2 1 1 1 0 2	Cu 81 76 68 61 69 77	Ga 20 17 14 14 16 17	Ge 0 1 0 0	La 16 17 12 14 19
SAMPLE 04-2217 04-2218 04-2219 04-2220 04-2221 04-2222	Mn 834 801 653 608 629 729	Mo 1 2 2 2 2 0	Nb 5 5 4 5 8 5	Ni 224 212 173 163 171 203	Pb 17 15 11 11 17	Rb 37 35 28 24 24 24 34	Sb 1 1 0 6 3 3 3	Sc 23 21 18 18 20 22	Sr 46 49 44 41 40 41	V 293 246 254 269 353 250	W 0 3 1 0 2 0	Y 15 13 14 14 12	Zn 70 66 51 46 46	Zr 123 124 140 118 120 123





SAMPLE	EASTING	NORTHING	SiO2	1102	AI203	Na20	Fe203	S	MgO	CoO	P205	K20	Ag	As	Au(ppb)	В	Ва	
108083	2690	4200	75.6	0.11	5.1	0.38	14.5	0.008	0.05	0.02	< 0.1	0.30	0.10	190	3	100	71	
108085	2675	4200	78.6	0.73	11.6	0.61	0.5	0.2112	0.47	0.04	< 0.1	2.89	0.40	35	69	500	520	
108087	2660	4200	93.4	0.14	3.5	0.49	0.3	0.0272	0.11	0.06	<0.1	1.00	0.10	16	42	40	230	
108090A	2645	4200	27.3	0.68	9.1	0.42	52.1	0.008	0.23	0.07	<0.1	0.54	0.10	2920	54	150	110	
108090B	2645	4200	51.9	1.61	25.7	2.21	2.9	0.0448	0.47	0.10	< 0.1	1.72	0.10	74	28	80	460	
108092A	2630	4200	32.0	0.77	8.7	0.84	45.6	0.016	0.25	0.02	< 0.1	80.0	0.10	220	26	150	9	
108092B	2630	4200	64.3	1.25	16.3	2.74	1.2	0.0528	0.80	0.11	< 0.1	0.33	0.20	26	110	100	44	
108094	2616	4200	63.6	0.43	20.8	0.11	0.6	0.0832	0.88	0.02	<0.1	5.02	0.10	41	10	100	610	
108096A	2600	4200	48.5	0.30	10.7	0.80	25.7	0.0912	0.82	0.13	<0.1	3.08	0.20	1910	3	150	300	
108096B	2600	4200	57.1	0.60	20.9	1.60	2.5	0.3648	1.31	0.06	<0.1	6.40	0.20	120	14	100	640	
108098A	2587	4200	28.9	0.49	6.9	0.70	51.8	800.0	0.32	0.02	<0.1	0.82	0.10	1540	340	150	150	
1080988	2587	4200	59.3	1.48	17.9	3.59	1.6	0.0608	0.74	0.18	<0.1	2.30	0.10	38	79	150	440	
108101A	2560	4200	19.0	1.18	11.8	0.59	54.7	0.008	0.40	0.10	<0.1	2.03	0.20	4980	3	200	340	
108101B	2560	4200	45.2	2.39	28.8	2.84	2.3	0.032	0.76	0.05	<0.1	4.34	0.20	43	65	120	840	
108103	2555	4200	9.3	0.69	7.0	0.24	69.3	0.0192	0.33	80.0	0.44	1.27	0.10	1430	160	120	210	
108105A	2540	4200	23.2	0.86	9.6	0.39	54.5	0.008	0.38	0.14	< 0.1	1.83	0.30	830	950	1000	50	
108105B	2540	4200	55.8	1.76	11.3	1.19	9.9	800.0	5.98	0.09	<0.1	1.16	0.10	140	350	400	230	
108107A	2520	4200	9.5	0.92	3.9	0.23	73.1	0.008	0.53	0.08	0.26	0.12	0.20	2540	3	200	300	
108107B	2520	4200	30.4	1.08	9.9	0.24	2.1	3.424	1.85	16.30	<0.1	1.84	0.20	33	59	100	440	
108110	2460	4200	39.9	2.36	30.0	3.59	3.1	0.1024	1.19	0.39	< 0.1	3.82	0.20	140	84	150	570	
SAMPLE 108083 108085 108087	Co 49 3 3	Cr 130 85 21	Cu 150 20 3	Ga 30 40 10	Ge 5 5 4 6	Mn 155 155 155	Mo 10 2 1 8	Ni 67 <20 23 82	Pb <5 17 <5 <5 <5	Sb 7 3 1	Sc 8 32 3 28	\$n 1 20 2 2	Sr 21 130 20 23	87 110 20 560	W 8 6 5	Y 2 14 4 8	Zn 74 11 5	Zr 55 230 57 62
108090A	13	380	97	40	-	310		90	<5	2	49	3	28	220	19	20	95	130
108090B	3	150	14	40	2	155 .	2 3	45	<5	2	22	. 2	12	730	1	7	52	52
108092A	6	630	81	50 30	3	155	3	23	<5	3	19	3	13	49	13	6	8	82
108092B	5	65 360	18 17	30 40	3	155 155	1	50	56	2	18	15	50	18	9	8	12	150
108094 108096A	3 36	250 450	180	60	3	155	60	76	14	24	21	3	56	140	10	8	60	130
	36 11	93	29	60	3	155	3	27	16	1	23	5	130	92	14	7	27	170
108096B 108098A	9	480	83	50	3	465	8	51	<5	3	20	1	33	740	22	4	46	34
108098B	12	120	13	30	2	155	2	37	<5	2	53	2	64	290	49	8	10	86
108101A	46	540	130	40	6	387	10	190	9	2	48	2	36	840	40	14	560	100
108101B	13	210	26	50	2	155	i	300	<5	2	73	2	50	430	40	15	360	140
1081015	10	370	110	30	4	155	3	83	39	7	37	5	22	360	28	8	240	46
108105A	21	1940	240	40	4	387	8	86	<5	4	43	3	25	650	11	14	260	61
108105B	38	470	57	40	3	465	8	120	<5	2	34	8	32	240	17	18	150	150
108103B	36 17	860	99	80	5	1394	10	36	8	7	41	3	36	1350	8	5	690	55
108107A	9	160	18	25	2	155	2	50	5	2	39	2	75	180	13	6	280	70
108110	27	320	11	40	4	155	2	150	8	2	73	3	43	460	ĩ	14	360	140
100110	21	J20	"	10	7	733	2	130	v	•		•						
SAMPLE	Quartz	Hematite	Goethile	Maghemile	Muscovite	/ermiculite/ Chlorite	Koolinite	Smectile	Talc	Halite	Sulfates	Olhers						

SAMPLE	Quartz	Hematite	Goethile	Maghemile	Muscovite	/ermiculite/ Chlorite	Koolinite	Smectile	Talc	Halite	Sulfates	Others
108083	6	4	5		3	3	4	1	4	3	?Gypsum	
108085	6	,	•		5	4				4	6Aurile	
108087	6				5	3				4		
108090A	6	6	4	1	3	5		1	4			
108090B	6	٧	À		Š	6	6	·	6	6		
108092A	6	5	4	1	,	2	4		1	3	?Gycsum	
	-	,	7		7	5	6	3	6	6	. 0,000	Paragonite
108092B	6				,	5	U	J		-	?Gypsum	, or agorate
108094	0	(_		0	5			O	,	: Uyusu	D'4-
108096A	6	4	5		4	6		1		3		Paragonite
108096B	6				6	2				6	6Aiurite	
108098A	6	5	4	1	3	2	3	1	4			
108098B	6				6		6	3	6	6	?Gypsum	Paragonite, SPtagicciase
108101A	6	6	4	1	5	3	4		2	3	3Gypsum	
108101B	6	-			6	5	6		5	6		
108103	6	5	6	1	4	5			6			
108105A	6	5	5	2	3	3		1	3			
108105R	6	1	3	1	5	6	5	6	6			4Plagiaclase
	-	,	,	;		-	4	2	-			
108107A	6	6	3	1	2	5	4	2	0		c ·	
108107B	6				6		4	6	3	_	60yssum	
108110	6		3		6	3	6	1	4	6	40yesum	

Appendix 2: Tabulated and Graphed data for surficial rock samples along 4200N. The mineralogical key is as follows: 6=major, 5=abundant, 4=moderate, 3=significant, 2=minor and 1=trace. Bi, TI <3 ppm.

SAMPLE	EASTING	NORTHING	SiO2	TiO2	A1203	No20	Fe203	S	MgO	CoO	P205	K20	Ag	As	Au(ppb)	В	Ba
			52.2	0.73	14.8	0.28	8.8	0.02	4.97	2.16	<0.1	1.13	0.05	46	36	50	300
108078	2775	4200								-							
108079	2750	4200	64.0	0.61	11.2	0.23	9.8	0.02	3.19	0.98	<0.1	0.79	0.05	53	23	40	310
108080	2730	4200	69.1	0.60	10.1	0.31	8.7	0.02	1.73	0.49	<0.1	0.91	0.05	63	13	40	290
180801	2710	4200	72.0	0.51	8.4	0.27	10.3	0.02	1.14	0.63	<0.1	0.72	0.05	95	22	40	290
	2690	4200	66.6	0.63	10.9	0.41	9.2	0.02	1.70	0.95	<0.1	0.82	0.05	91	45	40	310
108082																40	
108084	2675	4200	68.4	0.54	8.0	0.25	13.6	0.02	1.17	0.78	<0.1	0.57	0.05	180	28	-	390
108086	2660	4200	66.7	0.60	10.2	0.26	8.8	0.02	1.92	1.83	<0.1	0.77	0.05	91	32	50	320
108088	2650	4200	59.7	0.52	11.9	0.25	6.8	0.02	2.54	3.90	< 0.1	0.97	0.05	66	47	60	330
108089	2643	4200	63.6	0.44	9.8	0.29	6.9	0.02	2.19	4.27	<0.1	0.77	0.05	80	41	50	310
													0.05	87	33	50	330
108091	2630	4200	65.7	0.58	10.0	0.28	9.3	0.02	1.81	1.72	<0.1	0.82					
108093	2616	4200	63.5	0.59	10.3	0.63	7.6	0.02	2.24	3.12	<0.1	0.94	0.05	62	42	60	320
108095	2600	4200	67.4	0.57	9.3	0.50	7.8	0.02	2.05	2.98	< 0.1	0.83	0.05	72	33	40	280
108097	2587	4200	75.1	0.57	5.9	0.35	11.3	0.02	0.93	1.21	< 0.1	0.53	0.05	120	19	50	300
108099	2571	4200	67.5	0.50	7.8	0.39	9.9	0.02	1.67	3.42	<0.1	0.63	0.05	84	28	40	290
												0.50	0.05	100	26	40	440
108100	2560	4200	64.5	0.53	7.0	0.17	15.3	0.02	1.34	3.40	<0.1						
108104	2540	4200	60.9	0.68	5.5	0.31	27.8	0.02	0.54	0.21	<0.1	0.30	0.05	160	24	50	420
108106	2520	4200	63.8	0.61	9.8	0.62	15.7	0.02	1.07	0.15	· <0.1	0.88	0.05	160	24	80	380
108108	2482	4200	66.8	0.79	9.1	0.58	16.0	0.02	0.62	0.10	<0.1	0.66	0.05	85	13	60	300
				0.45	8.4	2.25	11.7	0.056	3.66	3.33	<0.1	0.47	0.05	85	51	70	600
108109	2460	4200	56.1												23	60	360
108111	2440	4200	61.7	0.62	10.4	0.82	15.0	0.02	1.58	0.17	<0.1	0.71	0.05	97			
108112	2420	4200	68.6	0.62	7.4	0.27	16.7	0.02	1.03	0.20	<0.1	0.50	0.05	110	16	35	280
108113	2400	4200	58.2	0.56	8.0	0.37	15.8	0.02	1.10	0.18	<0.1	0.57	0.05	100	11	40	330
	_	_	_	_	_				~	C L	r.	c-	Sr	٧	w	Y	Zn
SAMPLE	Co	Cr	Cu	Ga	Ge	Mn	Mo	Ni	Pb	Sb	Sc	Sn					
108078	43	780	77	20	1	852	6	380	<50	ţ	10	3	67	150	5	13	83
108079	32	750	62	10	0.3	620	5	270	<50	ı	7	1	48	160	5	11	68
108080	22	570	50	10	1	465	3	170	<50	1	9	3	44	150	3	11	60
				-			5	-	<50	2	7	5	38	170	3	10	57
108081	18	600	51	10	0.3	465	_	130						-	5		
108082	22	520	54	10	0.5	620	4	150	<50	2	9	3	47	160	- 1	11	64
108084	19	800	66	15	0.5	387	8	140	<50	2	15	4	41	240	5	10	56
108086	22	500	57	10	0.5	465	3	150	<50	1	20	3	50	150	5	10	57
108088	23	400	61	15	0.8	465	3	160	<50	1	17	3	65	130	3	11	55
										i	9	2	63	120	4	10	49
108089	20	380	58	10	0.3	310	3	140	<50	!							
108091	20	560	54	10	0.3	465	2	150	<50	1	8	3	48	160	1	10	54
108093	22	440	57	10	0.3	620	2	150	<50	1	17	I	59	130	4	10	51
108095	19	430	51	10	1	620	3	140	<50	1	10	2	54	130	2	10	46
108097	14	640	43	10	0.4	465	3	96	<50	2	13	2	40	190	7	8	43
										i	8	ī	52	180	5	10	45
108099	20	580	52	8	0.2	465	3	130	<50	•					g	10	
108100	20	970	58	10	0.3	465	3	130	<50	2	7	4	51	230	•		52
108104	21	1760	56	15	0.4	697	4	130	<50	4	8	5	34	450	12	9	55
108106	19	830	56	15	0.4	620	3	130	<50	3	25	3	34	260	10	10	84
108108	18	970	47	10	0.3	620	3	120	<50	2	26	3	36	260	6	9	47
					0.3	387	5	140	<50	2	21	5	67	220	ġ	9	43
108109	18	760	60	15								-			4	10	57
108111	23	960	60	15	0.3	775	3	180	<50	2	17	3	36	260			
108112	19	1130	49	12	2	697	3	160	<50	2	7	5	32	230	8	9	46
108113	20	1070	50	12	0.3	620	3	160	<50	3	22	4	34	260	6	9	49
7007.10																	
									W # *	C	T1-	p. an.	Olhara				
SAMPLE	Quartz	Plagioclase	Hemalite	Goelhile	Maghemile	Calcite	Muscovite	/ermiculite/	Kaalinite	Smecliles	Talc	Rutile	Others				
		-						Chlorite									
108078	6	4	3	4	1	5	5	6	6	1	6						
		7	3	3	i	3	3	•	5	3	5	3					
108079	6	_	-	J						J		,					
108080	6	6	3		3	3	5	3	5		5						
108081	6	3	3	3	4	3	2		4		3	I					
108082	6	2	3	2	2	3	3	2	4	3	2	2					
108084	6	4	4	3	2	3	1	1	4		2	2	Paragonite				
108086	6	3	3	4	4	5	4		5	4	4	3	Paragonile				
		_	•	3	,	-	3		4	•	3	•					
108088	6	2	3	3	!	6	3	2	4	2	3		Paragonile				
		1	7														

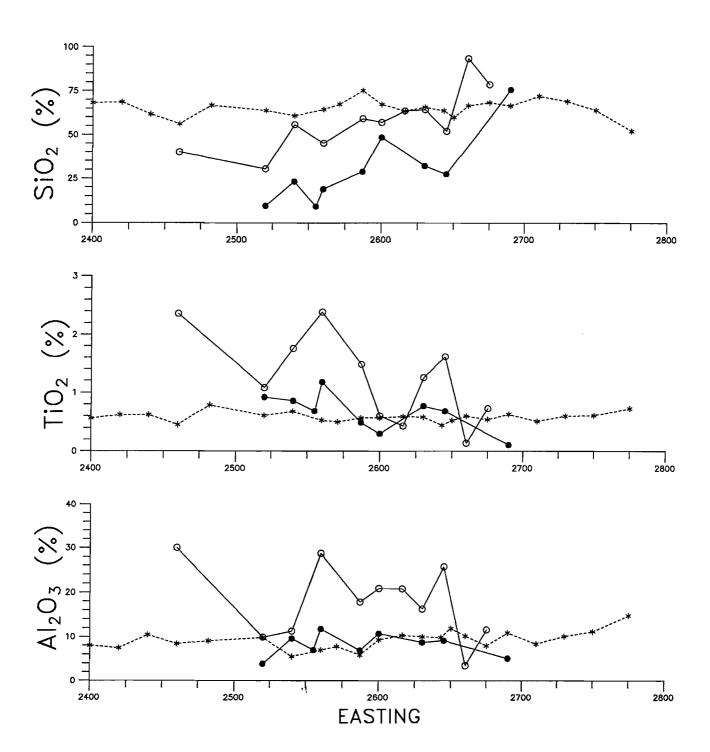
Paragonile

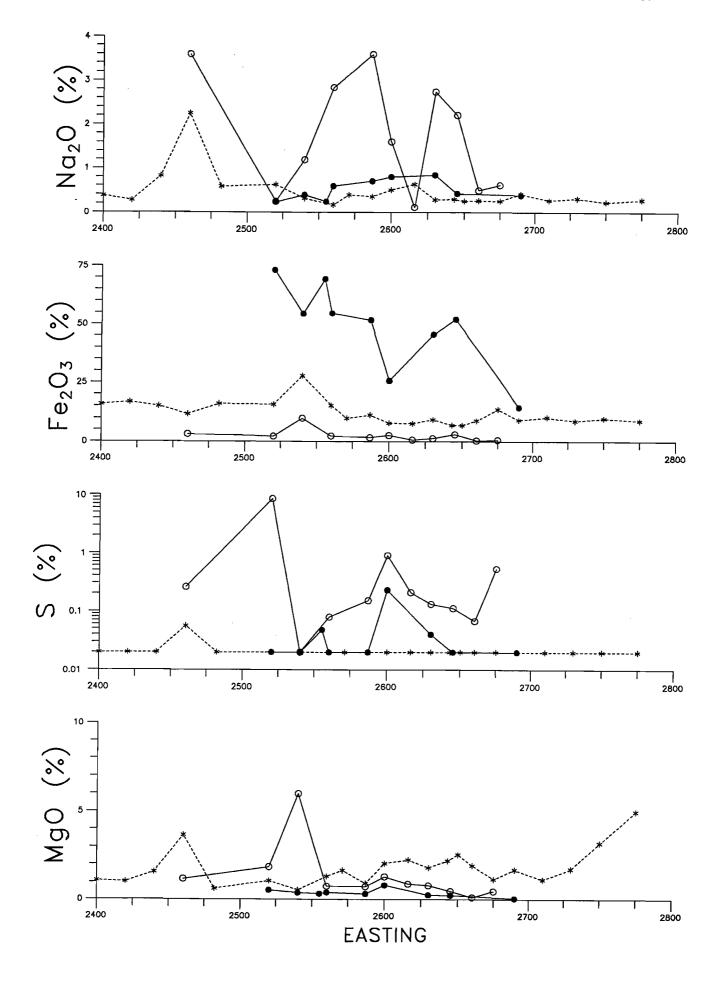
Paragonite 'IAmphibole Paragonite

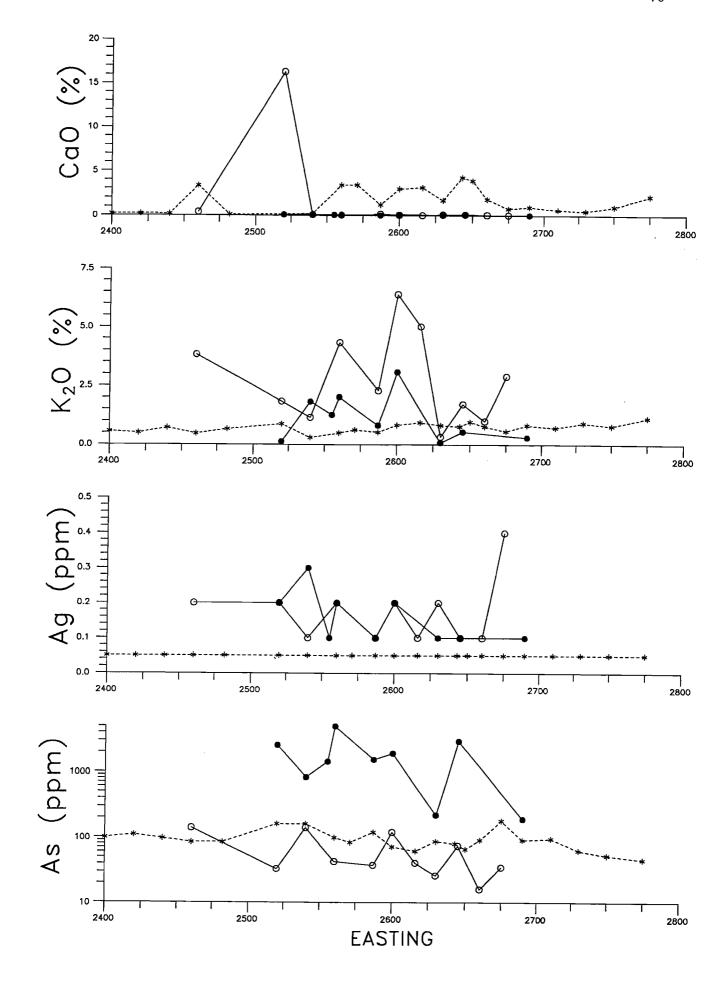
Appendix 2: Tabulated and Graphed data for surficial soil samples along 4200N. The mineralogical key is as follows: 6=major, 5=abundant, 4=moderate, 3=significant, 2=minor and 1=trace. Bi, Tl <3 ppm.

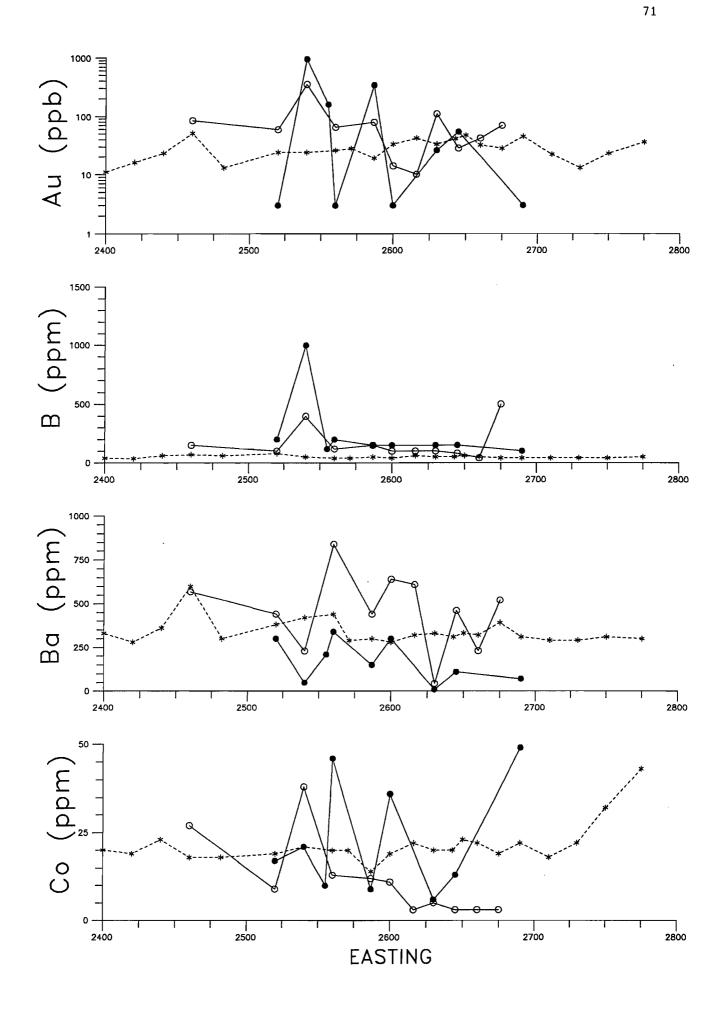
2. Near—surface samples: Graphed data for 4200N

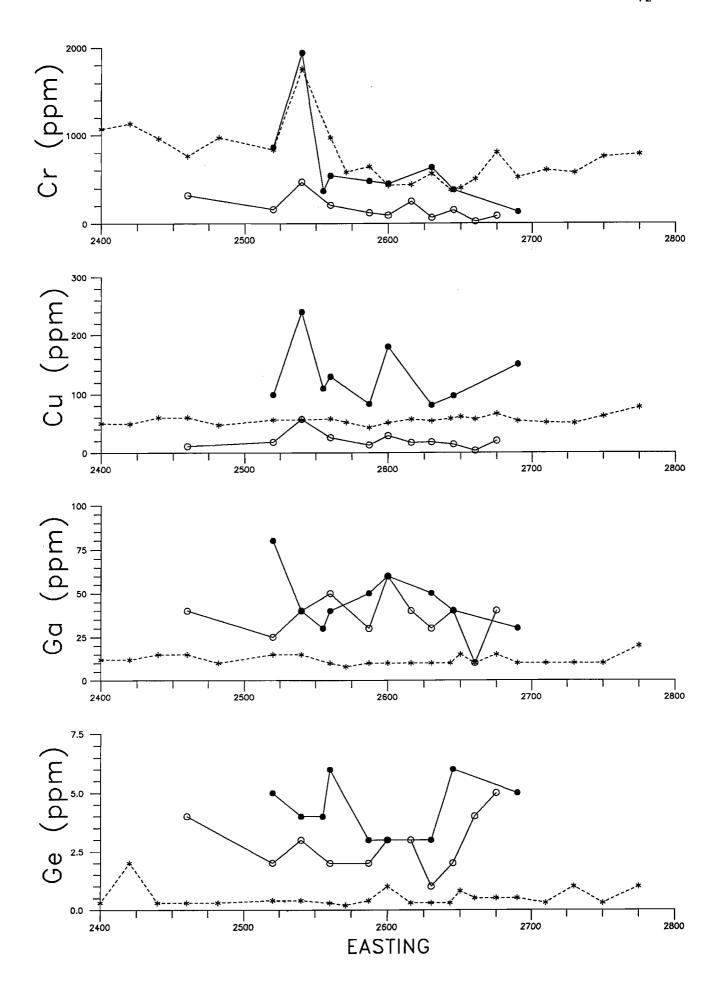
***** Soil
•••• Fe-rich weathered rock
•••• Fe-poor weathered rock

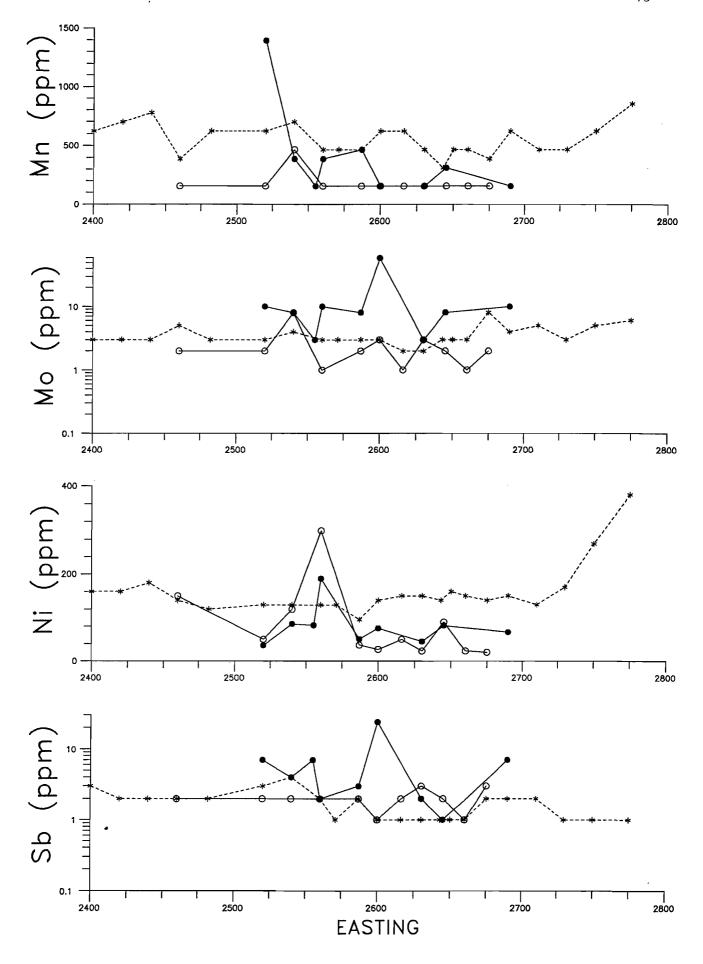


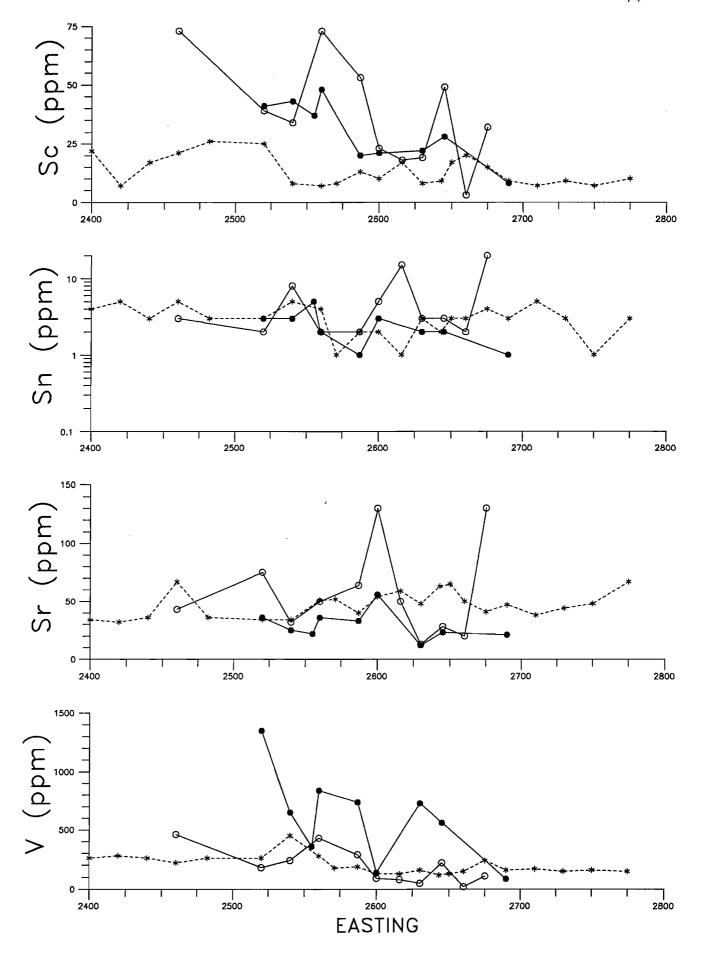


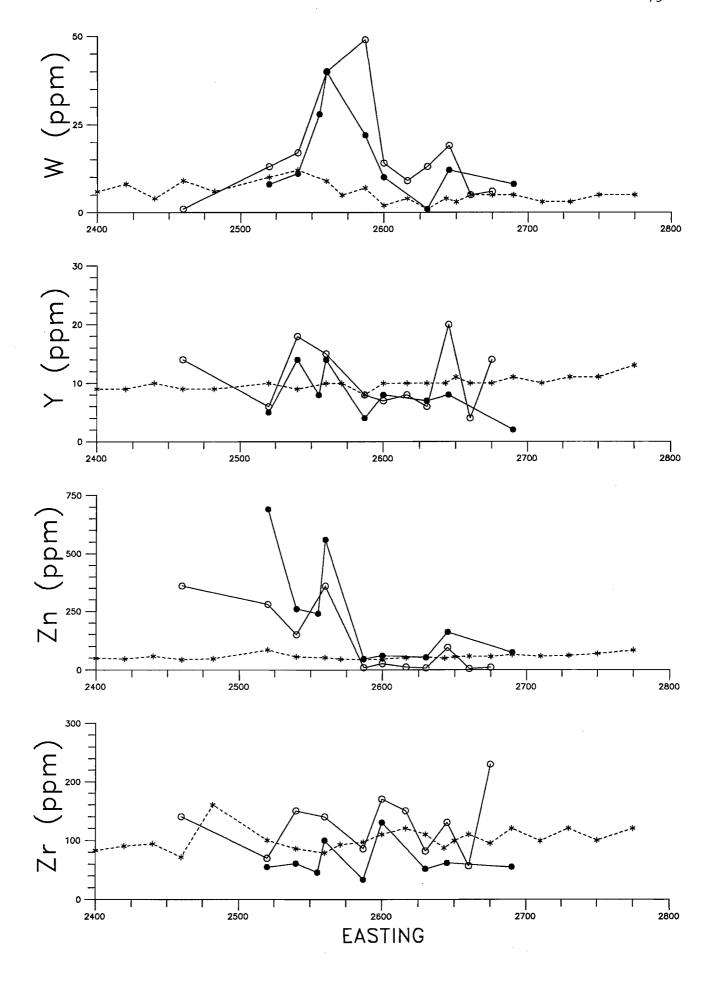


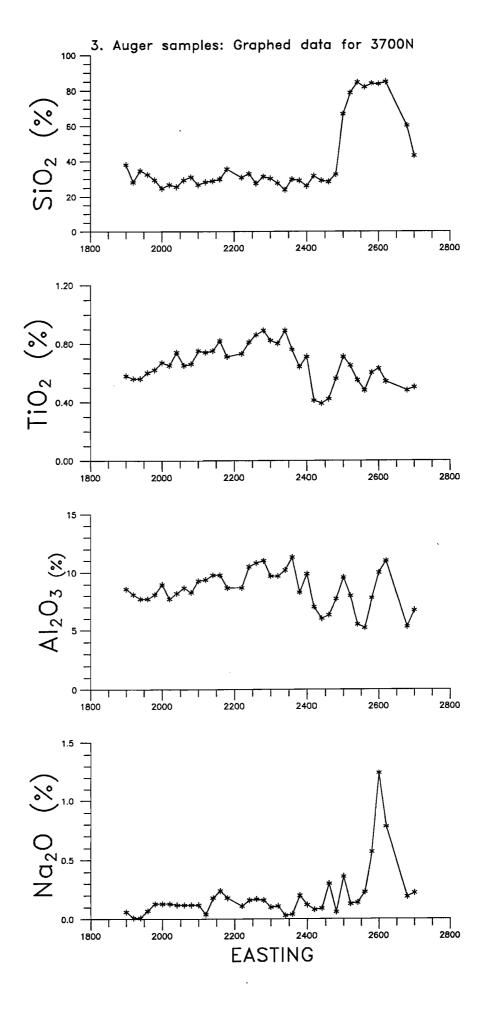


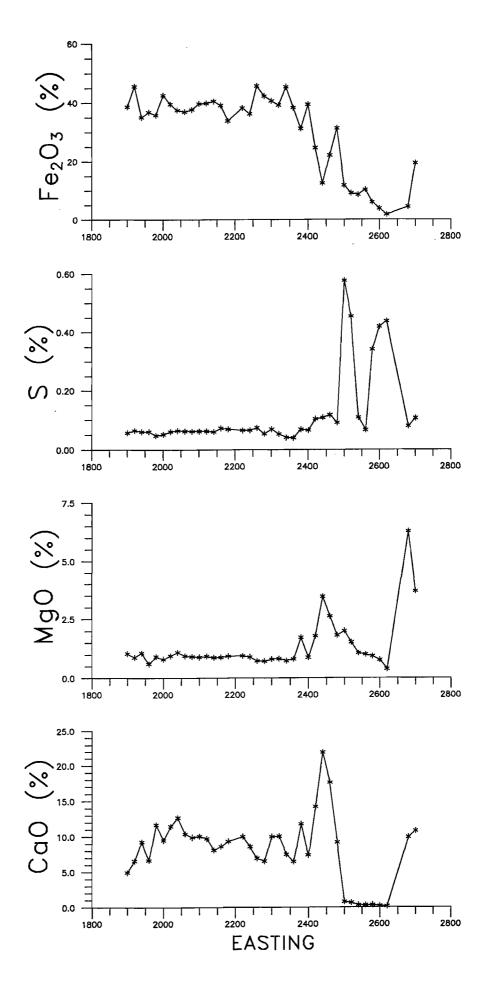


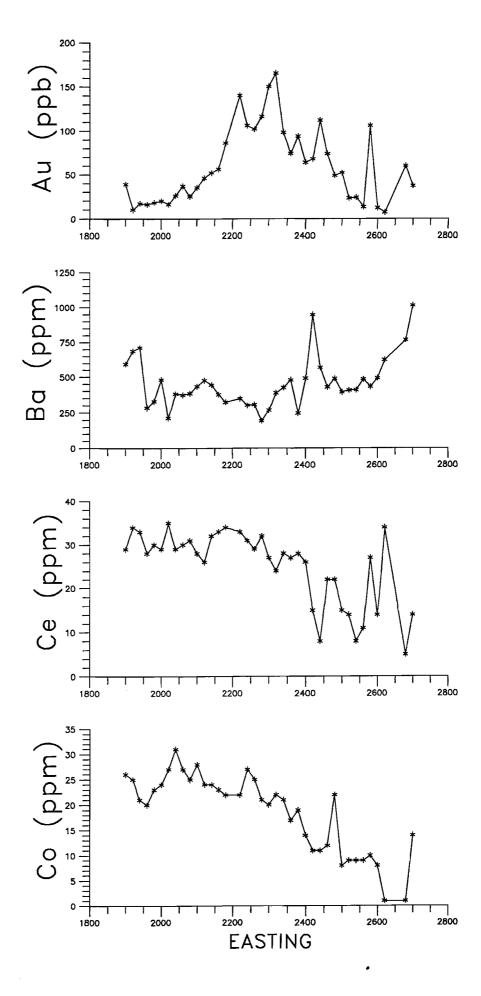


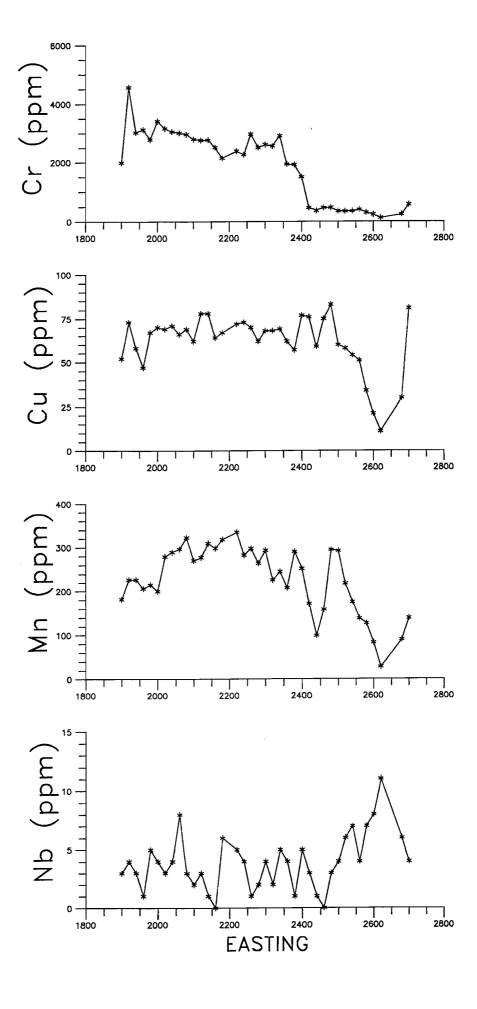


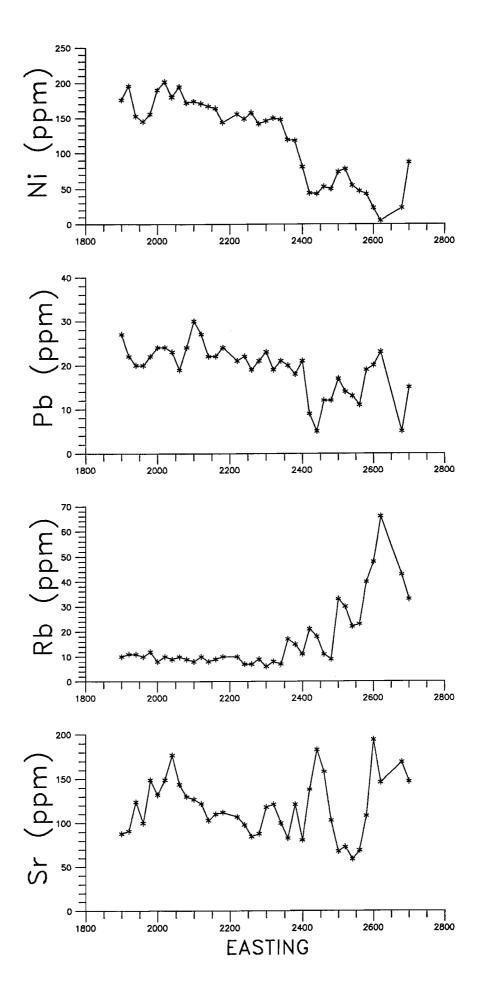


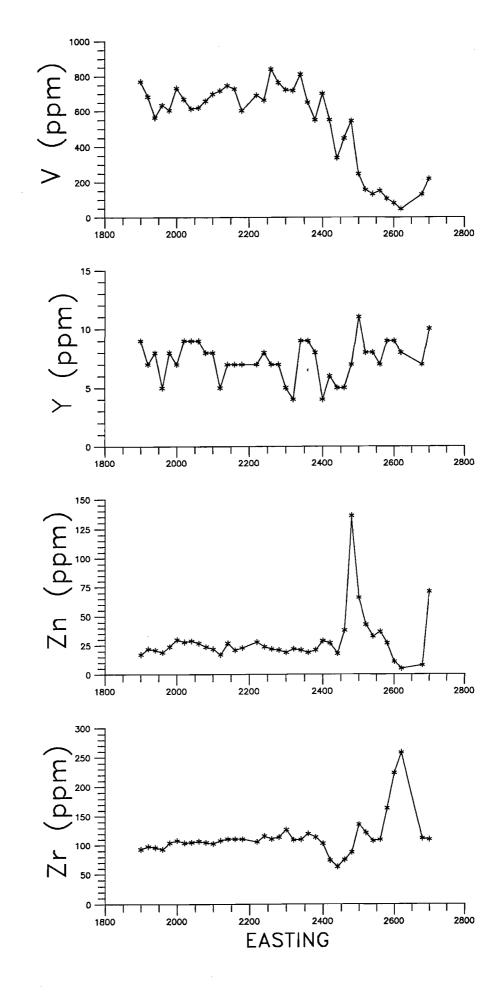












EASTING					EASTING				
FROM	TO	SAMPLE	Au (ppb)	As(ppm)	FROM	TO	SAMPLE	Au (ppb)	As(ppm)
2232	2234	91564	20	250	2406	2408	42353	10	130
2238	2240	91568	20	210	2408	2410	42354	170	120
2242	2244	91570	10	260	2410	2412	42355	10	190
2246	2248	91572	9	260	2412	2414	42356	10	75
2250	2252	91574	2	260	2414	2416	42357	10	70
2252	2254	91575	5	240	2416	2418	42358	10	110
2256	2258	91578	10	200	2418	2420	42359	60	85
2260	2262	91580	8	240	2420	2422	42360	20	60
2264	2266	91582	20	230	2422	2424	42361	60	90
2268	2270	91584	2	250	2424	2426	42362	10	65
2272	2274	91586	2	250	2426	2428	42363	80	50
2276	2278	91588	2	260	2428	2430	42364	910	160
2280	2282	91590	2 2	270	2430	2432	42365	20	55
2284	2286	91592	5	390	2432	2434	42366	10	90
2288	2290	91594	2	340	2434	2436	42367	60	75
2292	2294	91596	2	270	2436	2438	42368	40	95
2296	2298	91598	2	230	2438	2440	42369	10	90
2300	2302	91600	2	250	2440	2442	42370	20	6û
2302	2304	91601	5	280	2442	2444	42371	40	90
2306	2308	91604	2 7	390	2444	2446	42372	80	140
2310	2312	91606	20	480	2446	2448	42373	20	130
2314	2316	91608	7	530	2448	2450	42374	80	35
2320	2322	91612	10	490	2450	2452	42375	20	65
2324	2326	91614	30	620	2452	2454	42376	80	90
2328	2330	91616	10	650	2454	2456	42377	10	65
2332	2334	91618	20 -	260	2456	2458	42378	10	90
2336	2338	91620	30	280	2458	2460	42379	20	75
2340	2342	91622	20	210	2460	2462	42380	10	80
2344	2346	91624	30	190	2462	2464	42381	40	65
2346	2348	91625	30	150	2464	2466	42382	10	50
2350	2352	91628	60	160	2466	2468	42383	10	50
2354	2356	91630	80	110	2468	2470	42384	10	110
2358	2360	91632	70	260	2470	2472	42385	10	130
2362	2364	91634	40	310	2472	2474	42386	60	35
2366	2368	91636	40	210	2474	2476	42387	60	40
2370	2372	91638	60	200	2476	2478	42388	10	7
2374	2376	91640	70	250	2478	2480	42389	10	50
2378	2380	91642	110	250	2480	2482	42390	130	60
2382	2384	91644	60	250	2482	2484	42391	60	55
2386	2388	91646	20	180	2484	2486	42392	10	40
2390	2392	91648	70	220	2486	2488	42393	10	80
2394	2396	91650	80	180	2490	2492	42395	10	20
2400	2402	42350	20	80	2494	2496	42397	10	4
2400	2402	42351	40	95	2498	2500	42399	10	45
2402	2404	42352	20	130	2502	2504	42401	10	100
2404	2400	72332	20	130	2302	2007	72701	10	100

Appendix 4: Tabulated data for the trench samples along 3700N (see Figure 13)

SAMPLE	SPECIES	EASTING	NORTHING	ASH (%)	Ca (%)	Fe (%)	Mg (%)	Na (%)	P (%)	A	As	Au	Ba
04-2001 04-2002	Atriplex sp. Atriplex sp.	3300 3273	4200 4200	22.4 17.7	3.08 4.76	0.13 0.18	1.12 1.32	29.00 24.89	0.36 0.32	2025 2051	<7 <6	0.018 0.002	65 65
04-2003 04-2004	Atriplex sp.	3246 3219	4200 4200	26.6 6.2	2.90 24.45	0.06 0.42	1.86 2.68	25.66 9.06	0.35 0.95	741 4720	5 15	0.012 0.132	19 42
04-2005	Melaleuca sp. Haiosarcia sp.	3192	4200	23.8	1.67	0.41	1.40	34.91	0.14	3725	<5	0.004	26 45
04-2006 04-2007	Melaleuca sp. Melaleuca sp.	3165 3138	4200 4200	7.2 6.4	15.83 24.66	0.57 0.51	2.86 2.82	12.47 7.01	0.48 0.76	5838 5391	18 14	0.018 0.040	44
04-2008	Melaleuca sp.	3111	4200	8.0	24.80	0.35	1.27	7.37 4.56	0.38 0.38	4071 23162	29 51	0.029 0.005	32 58
04-2009 04-2010	Lawrencia sp. Melaleuca sp.	3084 3057	4200 4200	14.3 5.7	8.00 19.24	2.43 0.49	2.03 2.78	9.56 13.44	0.67	6473	19	0.038	68
04-2011	Halosarcia sp.	3005 2978	4200 4200	29.7 26.1	1.34 0.95	0.19 0.18	1.62 0.84	33.06 40.61	0.15 0.09	2363 2280	6 <5	0.008 100.0	21 20
04-2012 04-2013	Halosarcia sp. Dodonaea sp.	2951	4200	3.4	8.88	0.29	3.92	1.91	3.90	3310	<6	0.185	116
04-2014 04-2015	Atriplex sp. Halosarcia sp.	2924 2897	4200 4200	17.0 40.9	4.18 1.75	0.15 0.18	1.90 1.38	33.39 23.27	0.57 0.16	1899 2188	<5 <4	<0.001 <0.001	220 18
04-2016	Atriplex sp.	2870	4200	20.0	3.39	0.11	1.16	29.29	0.43	1367	<5 <5	0.002 0.016	76 52
04-2017 04-2018	Eremophila sp. Eremophila sp.	2843 2816	4200 4200	4.4 4.4	10.37 10.16	0.23 0.26	2.41 2.28	1.28 1.25	3.22 3.81	3558 3649	7	800.0	45
04-2019	Atriplex sp.	2789	4200	21.8 4.7	3.14 10.94	0.09 0.20	1.31 1.65	26.48 0.68	0.33 5.04	1392 2727	<4 <6	0.015 0.009	74 38
04-2020 04-2021	Eremophila sp. Alriplex sp.	2767 2735	4200 4200	18.5	3.31	0.12	1.23	25.87	0.21	1830	<5	0.003	53
04-2022 04-2023	Atriplex sp. Eremophila sp.	2708 2681	4200 4200	20.3 5.0	3.66 8.37	0.13 0.21	1.23 1.36	23.00 0.38	0.26 1.51	2404 3200	<4 10	0.004 0.014	57 49
04-2024	Atriplex sp.	2654	4200	22.9	2.56	0.08	1.13	28.52	0.22	1352	<4	0.003	75
04-2025 04-2026	Halosarcia sp. Eremophila sp.	2613 2590	4200 4200	33.7 4.5	0.78 12.85	0.15 0.26	0.60 1.75	35.04 1.71	0.18 2.08	1981 3362	8 24	<0.001 0.016	18 65
04-2027	Atriplex sp.	2558	4200	23.9	2.51	80.0	1.00	24.18	0.22	925	<4 18	<0.001 0.023	71 83
04-2028 04-2029	Eremophila sp. Eremophila sp.	2536 2479	4200 4200	4.5 4.5	8.58 9.18	0.44 0.25	1.61 2.12	1.59 1.27	2.33 2.13	3893 2914	11	0.017	96
04-2030	Atriplex sp.	2447	4200	26.6	1.78 15.76	0.07 0.26	0.81 2.05	31.08 0.92	0.28 2.54	966 2624	<5 12	0.001 0.024	41 181
04-2031 04-2032	Eremophila sp. Atriplex sp.	2425 2393	4200 4200	4.3 27.8	2.14	0.07	0.79	28.31	0.32	780	<5	< 0.001	51
04-2033 04-2034	Eremophila sp. Atriplex sp.	2366 2339	4200 4200	4.5 20.8	13.41 3.18	0.37 0.08	2.43 1.41	0.94 28.39	1.42 0.24	2900 1077	11 <5	<0.001 <0.001	96 45
04-2035	Eremophila sp.	2307	4200	4.2	10.14	0.20	2.61	0.47	3.29	1600	<7	0.002	64
04-2036 04-2037	Atriplex sp. Eremophila sp.	2285 2253	4200 4200	22.7 4.1	2.20 12.12	0.06 0.46	1.20 2.47	36.36 1.10	0.36 2.66	726 5315	<5 <7	<0.001 0.014	52 104
04-2038	Alriplex sp.	2226	4200	26.2	2.51	0.08	0.73	36.04	0.20 2.40	1113 2796	<5 <8	<0.001 0.024	59 58
04-2039	Eremophila sp.	2200	4200	3.9	9.08	0.21	1.87	0.98					
04-2040 04-2041	Eucalyptus sp. Eremophila sp.	1925 1940	3700 3700	4.5 4.7	13.03 8.03	0.24 0.15	2.84 1.53	8.34 0.72	1.90 1.45	1481 1077	<7 <6	<0.001 0.005	295 57
04-2042	Eucalyptus sp.	1950	3700	5.8	10.75	0.16	2.35	9.06	1.28	1201	<7	< 0.001	216
04-2043 04-2044	Eremophila sp. Eucalyplus sp.	1980 2000	3700 3700	4.5 5.0	11.39 14.42	0.23 0.18	2.65 2.57	0.46 8.20	1,34 1,88	2083 1418	<9 6	0.007 0.003	31 274
04-2045	Eremophila sp.	2010	3700	4.4	9.84	0.18	1.76	0.76	1.55	1667	9	<0.001	42 40
04-2046 04-2047	Eremophila sp. Eucolyptus sp.	2038 2060	3700 3700	4.0 3.1	8.46 9.38	0.22 0.31	1.18 3.32	0.68 12.89	1.56 2.35	2166 2763	<8 <13	100.0> 100.0>	238
04-2048	Eremophila sp.	2075	3700	4.4	11.52	0.34	2.52 · 2.07	0.91 6.76	1.32 1.65	3152 1367	<9 <6	<0.001 <0.001	51 147
04-2049 04-2050	Eucolyptus sp. Eremophila sp.	2100 2133	3700 3700	5.2 4.9	11.00 15.80	0.21 0.20	2.97	0.82	1.39	1846	<8	< 0.001	48
04-2051 04-2052	Eucolyptus sp. Eremophila sp.	2150 2175	3700 3700	5.0 4.9	14.57 13.87	0.23 0.17	2.17 2.48	8.04 0.38	1.37 2.59	1924 1509	10 <8	<0.001 <0.001	235 60
04-2053	Eucolyplus sp.	2220	3700	4.2	18.79	0.35	2.20	6.19	1.63	3079	12	0.003	304 62
04-2054 04-2055	Eremophila sp. Eremophila sp.	2250 2275	3700 3700	4.7 4.2	11.33 11.19	0.34 0.28	2.09 1.93	0.89 1.26	1.45 1.61	4079 3679	6 9	0.029 0.011	53
04-2056	Eucalyptus sp.	2308	3700	4.8	15.31	0.27	2.71	9.58 10.92	1.20 1.03	3108 2682	<6 8	0.003 0.002	465 334
04-2057 04-2058	Eucalyplus sp. Eremophila sp.	2330 2345	3700 3700	5.0 4.7	14.29 13.62	0.31 0.25	2.29 1.85	0.65	1.08	2394	6	0.009	60
04-2059 04-2060	Eucalyptus sp. Eremophila sp.	2375 2400	3700 3700	4.8 5.2	10.51 10.45	0.32 0.22	2.90 1.61	9.33 0.90	1.46 1.36	2253 2224	8 12	0.007 0.016	173 59
04-2061	Eucolyptus sp.	2425	3700	3.9	12.01	0.38	3.16	11.20	1.18	3069	5 12	0.001 0.049	228 53
04-2062 04-2063	Eremophila sp. Eremophila sp.	2450 2475	3700 3700	5.1 4.4	12.98 16.77	0.26 0.24	1.91 2.99	0.59 1.25	1.34 1.47	2676 2371	14	0.059	53
04~2064	Eremophila sp.	2500	3700	5.3	18.87	0.25	1.74	1.13 24.70	2.61 0.33	3634 1613	11 <4	0.007 <0.001	68 65
04-2065 04-2066	Atriplex sp. Melaleuca sp.	2520 2550	3700 3700	16.2 7.1	3.80 31.49	0.14 0.33	1.27 2.77	3.91	0.96	3707	10	0.012	58
04-2067	Melaleuca sp.	2575	3700 3700	8.5 31.3	22.36 2.21	0.18 0.11	1.79 1.46	4.75 32.26	0.52 0.13	1998 1096	8 <3	0.009 <0.001	35 12
04-2068 04-2069	Halosarcia sp. Halosarcia sp.	2600 2625	3700	24.8	2.04	0.23	1.80	31.42	0.12	2094	<4	0.003	29
04-2070 04-2071	Halosarcia sp. Eucalyptus sp.	2650 2680	3700 3700	19.2 4.6	0.94 12.91	0.23 0.31	1.27 3.18	32.50 9.73	0.16 1.02	2967 1913	6 11	<0.001 0.009	29 277
04~2072	Eremophila sp.	2700	3700	4.4	12.16	0.27	1.96	0.77	1.40	2418 3955	18 12	<0.001 0.014	49 244
04~2073 04~2074	Eucalyplus sp. Eremophila sp.	2725 2750	3700 3700	5.0 4.6	11.11 10.65	0.39 0.46	2.18 1.74	12.69 1.07	1.19 1.40	4906	15	0.027	50
04~2075	Eucolyplus sp.	2775	3700 3700	4.7 4.9	12.08 13.63	0.38 0.24	2.70 1.71	9.10 1.09	1.66 1.42	2830 2291	6 16	0.006 0.010	121 38
04-2076 04-2077	Eremophila sp. Eucolyplus sp.	2805 2820	3700	4.0	13.63	0.48	2.73	14.04	0.86	4179	10	0.031	217
04-2078 04-2079	Eremophila sp. Eucalyptus sp.	2850 2875	3700 3700	4.4 5.0	8.76 12.07	0.23 0.35	2.03 1.45	0.43 10.94	1.48 1.05	2210 2790	<7 <6	0.002 0.017	64 234
04-2080	Eremaphila sp.	2905	3700	4.4	15.08	0.24	2.01	0.61	1.40	2148 2486	<7 6	0.017 0.009	84 47
04~2081 04~2082	Eremophila sp. Eucalyplus sp.	2925 2940	3700 3700	4.9 3.8	14.38 13.61	0.38 0.25	1,51 2.92	0.72 11.14	2.57 1.38	2751	11	0.042	345
04-2083	Atriplex sp.	2975	3700	5.7	19.38	0.40	6.99	0.71 0.67	0.96 1.21	4208 2841	7 9	0.019 0.003	761 38
04-2084	Eremophila sp.	3000	3700	4.5	14.03	0.28	1,71						
04-2130	Eucolyptus sp.	2280 2280	3775 3775	5.6 4.8	20.70 12.77	0.31 0.42	2.16 2.17	5.52 0.77	0.92 2.21	2779 4167	<5 8	0.009 0.026	194 42
04-2131 04-2132	Eremophila sp. Eucalyplus sp.	2300	3792	4.4	12.22	0.37	2.89	8.69	0.99	3255	14	810.0	191
04-2133 04-2135	Eremophila sp. Eremophila sp.	2300 2410	3792 38 40	5.6 4.6	6.09 6.99	0.17 0.26	1.21 1.51	0.35 0.55	1.53 1.39	1850 4008	5 17	<0.001 0.015	33 67
04-2137	Eremophila sp.	2408	3770	3.9	16.27	0.53	3.39	1.54	1.46	5492	15 11	0.037 0.003	50 121
04-2138 04-2139	Eucolyptus sp. Eremophila sp.	2375 2375	3725 3725	4.5 5.2	7.73 11.73	0.28 0.24	2.32 1.68	14.08 0.64	0.99 1.38	2453 2506	10	0.005	38
				4.7	13.93	0.42	2.10	10.61	1.19	3145	<6	0.009	175
04-2140 04-2141	Eucalyptus sp. Eremophila sp.	2250 2250	4925 4925	5.2	14.78	0.75	1.51	1.03	1.61	6781	8	0.033	68
04-2142 04-2143	Eucolyptus sp.	2350 2350	4900 4900	5.0 5.8	13.81 18.11	-0.43 0.56	2.72 2.33	13.28 0.39	0.99 1.70	2664 4016	<6 12	0.015 0.039	167 81
				-									

Appendix 5: Tabulated and graphed data for the vegetation samples.

SAMPLE 04-2001 04-2002 04-2003 04-2005 04-2006 04-2009 04-2010 04-2010 04-2013 04-2013 04-2014 04-2014	Atripiex sp. Atripiex sp. Melaleuca sp. Halosarcia sp. Melaleuca sp. Melaleuca sp. Melaleuca sp. Lawrencia sp. Halosarcia sp. Halosarcia sp. Halosarcia sp. Dodonaea sp.	Be	Br 1662 1749 2010 486 453 229 192 163 137 391 539 281 158 393		Cr 20 21 9 67 25 65 78 46 257 54 17	Cu 57 69 41 108 73 191 157 98 153 185 46 88 317 73	Mn 476 418 95 652 206 1226 397 257 297 379 171 861 751	Ni 6 5 32 42 8 24 23 14 80 25 12 14 <1	96 41 41 41 41 41 41 42 41 41 41	\$i 247 679 129 858 252 817 578 459 255 1053 97 103 1786 166	Ti 103 101 26 222 272 247 189 145 542 305 185 191 183	V 6 7 3 14 16 13 17 12 83 16 7 7	# 41 41 41 41 41 41 41 41 41 41 41 41 41
04-2015 C4-2017 C4-2017 C4-2019 C4-2020 C4-2021 C4-2023 C4-2024 C4-2025 C4-2027 C4-2028 C4-2030	Atripiex sp. Eremophila sp. Eremophila sp. Eremophila sp. Atripiex sp. Eremophila sp.		445 493 574 708 476 850 670 851 897 676 573 796 687 692 938 843 537 994 813 657 1124 611 722 1077 767		23 17 34 38 14 31 15 16 15 8 10 39 11 49 33 41 146 13 36 70 13	47 40 380 454 48 422 45 48 270 44 37 436 25 376 480 328 28 281 27 331 30 379 34 325	120 285 484 508 396 520 298 1103 470 259 187 774 169 768 667 201 1134 364 578 181 502 216 466 261 515	8 <1 <1 <1 <1 <1 <1 <1 <1 <1 <1 <1 <1 <1	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	51 390 1550 1249 272 1309 308 219 1134 208 242 1700 915 1506 1241 477 2651 855 2743 733 5628 276 1497 263	116 73 145 157 53 107 70 87 108 52 141 106 44 159 114 39 105 30 100 44 72 33 180 46 106	7 4110 47 5603 69411 82637 33211 35	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
04-2040 04-2041 04-2042 04-2043 04-2046 04-2047 04-2050 04-2051 04-2053 04-2053 04-2054 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2056 04-2057 04-2058 04-2059 04-2059 04-2059 04-2059 04-2059 04-2060 04-2061 04-2060 04-2060 04-2061 04-2065 04-2065 04-2067 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2070 04-2080 04-2080 04-2080 04-2080 04-2080 04-2083 04-2084	Eucolyptus sp. Eremophila sp. Eucolyptus sp. Eremophila sp. Eucolyptus sp. Eremophila sp. Eucolyptus sp. Eremophila sp. Eucolyptus sp. Eremophila sp. Eucolyptus sp. Eremophila sp.		437 1227 432 1481 567 976 11678 499 1585 493 1400 602 741 919 422 344 1117 316 812 439 691 179 98 221 159 98 221 179 179 322 179 179 321 699 416 699 416 699 417 699 418 699 409 409 409 409 409 409 409 409 409 4	4 < 1 4 < 1 4 < 1 4 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1 5 < 1	77 34 64 44 53 42 58 68 36 68 36 54 42 55 42 42 42 42 42 42 43 42 42 43 44 42 43 44 44 45 46 46 47 47 47 47 47 47 47 47 47 47 47 47 47	131 500 132 310 112 376 412 234 603 149 360 107 383 129 444 306 137 210 256 163 402 154 390 44 47 390 44 47 391 402 154 402 154 391 402 154 403 170 170 170 170 170 170 170 170 170 170	416 278 260 505 485 294 462 536 562 328 412 664 1023 356 555 631 443 246 1372 661 1277 618 822 324 113 178 1100 1398 322 648 350 559 648 369 649 178 178 178 178 178 178 178 178 178 178	65 66 68 26 52 19 36 53 56 59 58 59 59 54 55 55 55 56 68 57 77 53 54 56 68 57 77 57 57 57 57 57 57 57 57 57 57 57	$\begin{array}{c} \bullet \bullet$	1449 1286 1260 3292 1180 4040 2060 1907 1805 1694 2248 1155 1628 1155 1628 1159 1184 1529 1184 1529 1322 2962 1574 1677 594 1112 700 318 1427 285 1732 1707 1369 1674 1843 1772 1772 1774 1843 1772 1772 1774 1772 1774 1772 1774 1774	73 67 60 112 68 85 101 163 139 73 64 91 49 149 150 1150 1150 1150 105 148 109 105 148 109 105 148 109 1157 75 83 109 109 1151 1151 1151 1151 1151 1151 1	53452357104464886787767767626471067115771166385106	6 < 1 < 2 < 1 < 1 < 2 < 1 < 1 < 2 < 1 < 1
04-2131 04-2132 04-2133 04-2135 04-2137 04-2138 04-2139	Eucalyptus sp. Eremophila sp. Eucalyptus sp. Eremophila sp. Eremophila sp. Eremophila sp. Eucalyptus sp. Eremophila sp. Eremophila sp.	<1 <1 <1 <1 <1 <1 <1	302 778 467 539 761 406 411	8 2 3 <1 7 5 7 3	65 58 85 27 39 70 38 38	113 417 145 448 383 290 196 241	428 625 649 561 346 520 2826 937	67 41 59 31 44 43 66 29	1 <1 17 <1 <1 9 17 <1	1100 1152 1580 2916 1488 1656 1847 1214	108 189 160 118 198 434 192 130	6 9 9 5 7 15 7 8	<1 <1 4 <1 <1 1 2
04-2140 04-2141 04-2142 04-2143	Eucalyptus sp. Eremophila sp. Eucalyptus sp. Eremophila sp.	<1 <1 <1	452 872 419 1107	9 14 <1 <1	123 128 103 78	205 376 130 278	500 651 586 531	100 87 92 60	12 <1 <1 8	1782 1482 1049 520	247 362 202 125	10 19 4 9	2 2 <1 <1

Appendix 5 (continued): Tabulated and graphed data for the vegetation samples.

5. Vegetation sample data: Graphed data for 3700N.

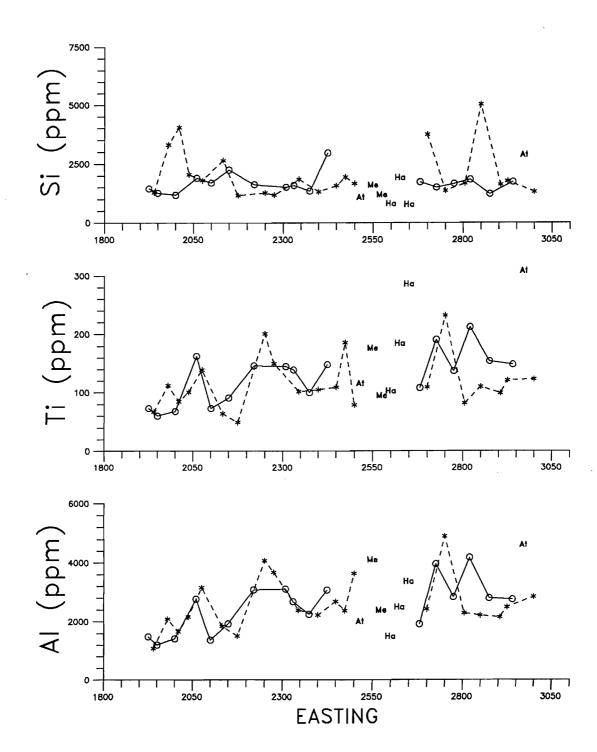
eeeee Eucalyptus spp.

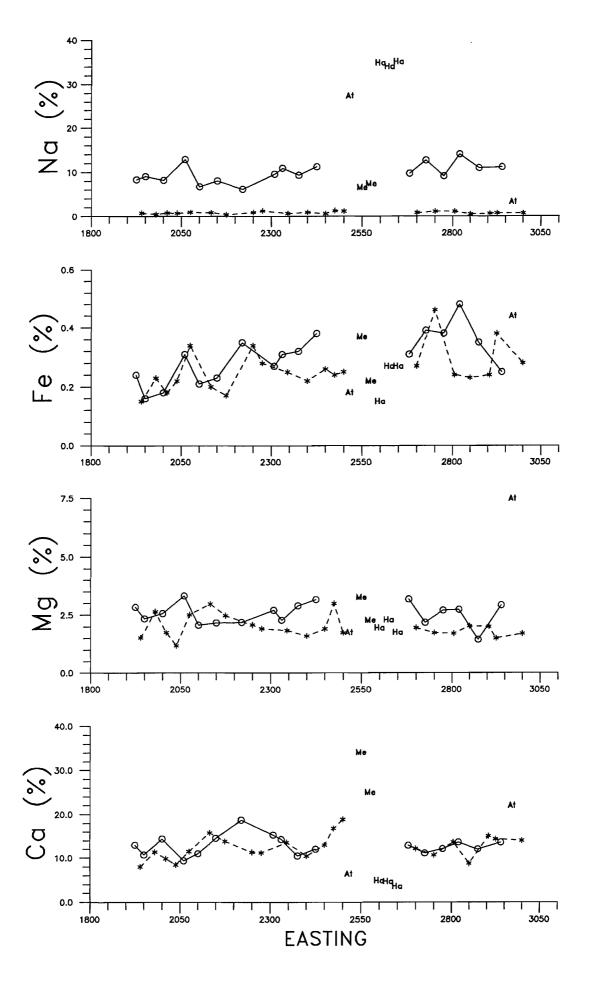
***** Eremophila spp.

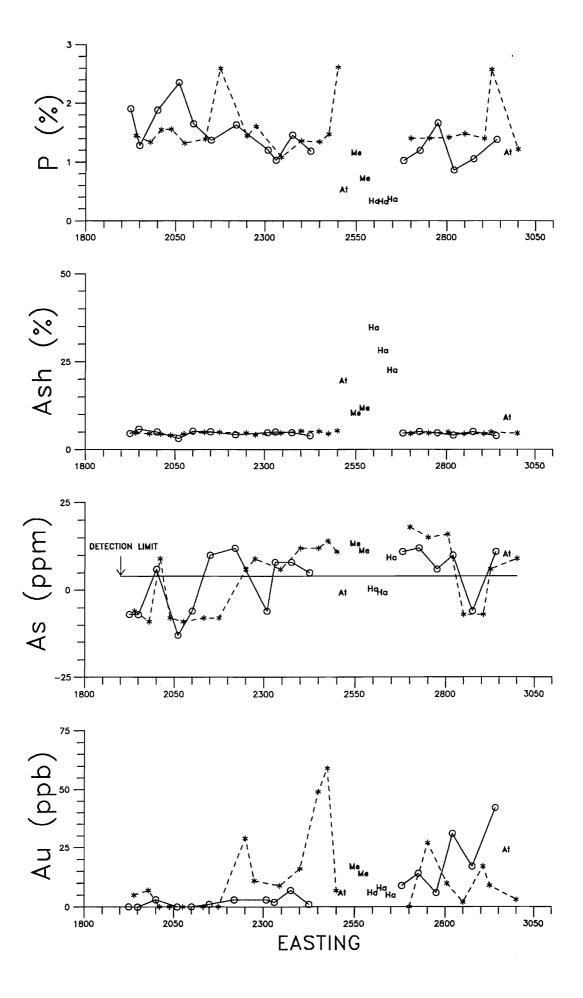
Me Melaleuca spp.

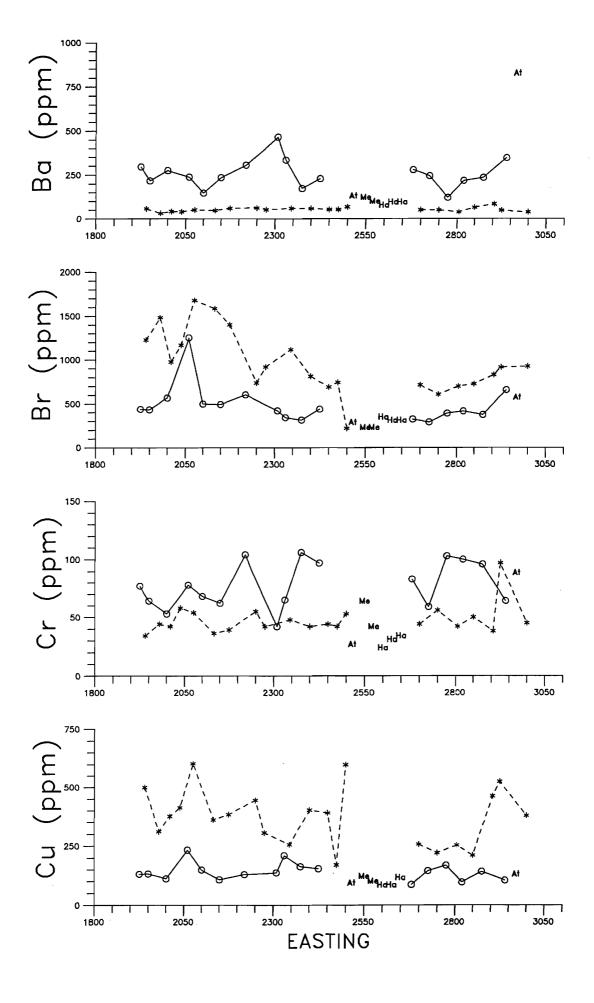
Ha Halosarcia spp.

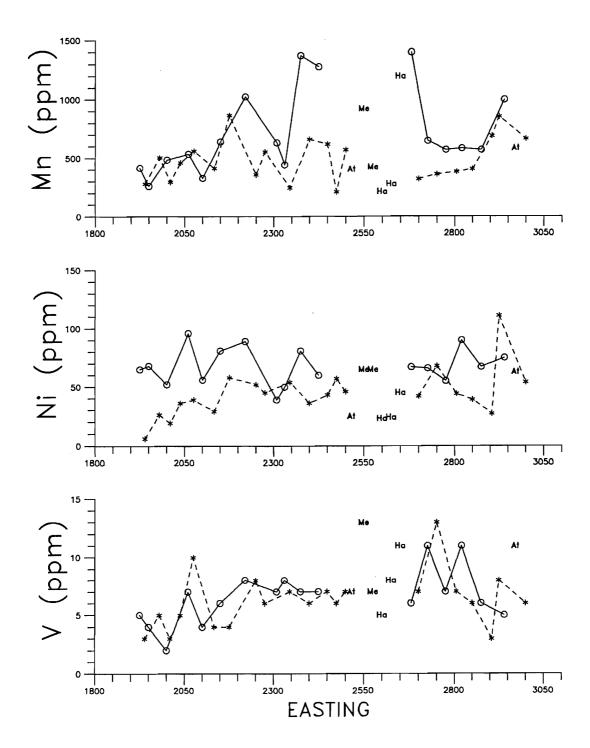
At Atriplex spp.











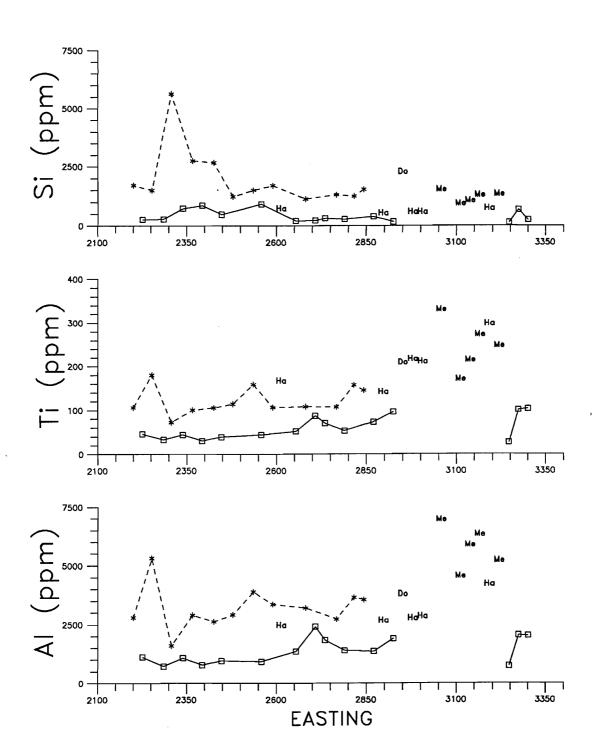
5. Vegetation sample data: Graphed data for 4200N.

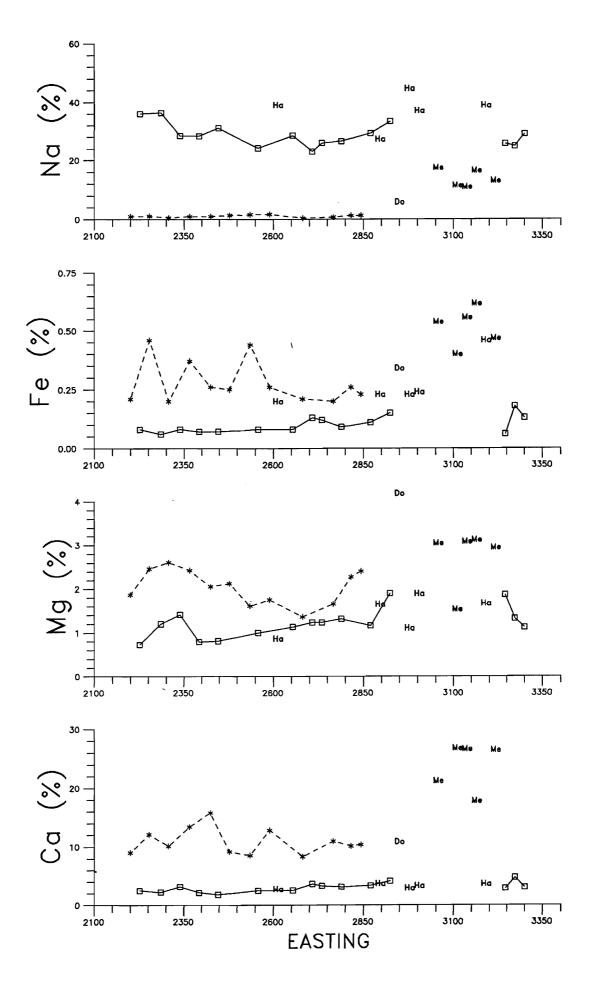
***** Eremophila spp.

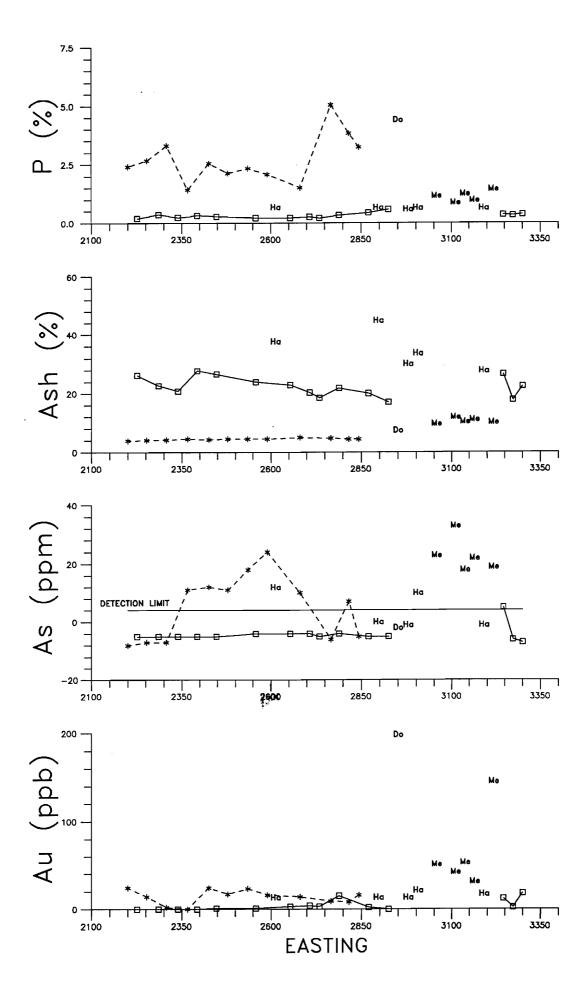
Me Melaleuca spp.

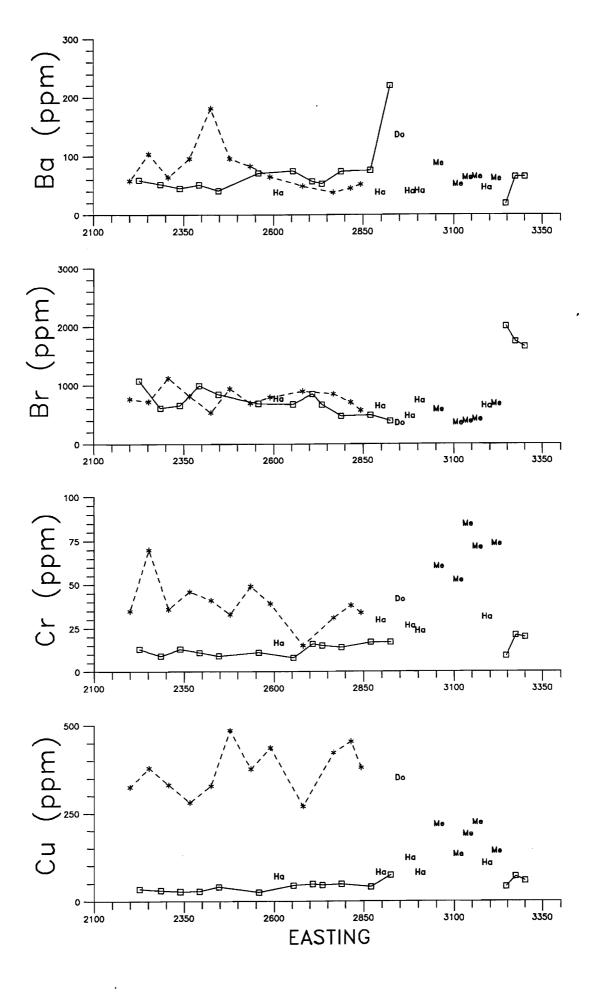
Ha Halosarcia spp.

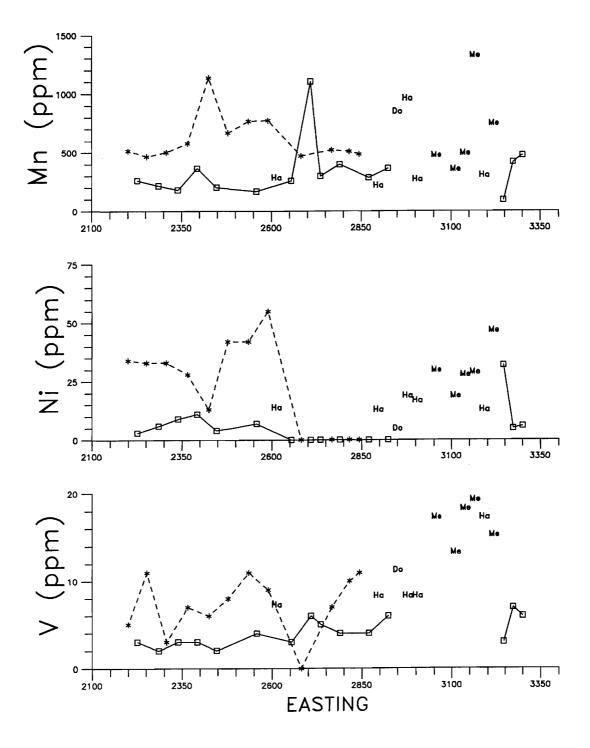
Do Dodonaea sp.











SAMPLE 04-2112 04-2129	TYPE Whole Whole	SiO2 27.30 36.60	Al203 9.40 11.29	Fe2O3 43.80 36.50	MgO 0.40 0.44	CaO 2.63 2.06	Na20 <0.01 0.22	TiO2 0.38 0.66	P205 < 0.01 < 0.01	Au(ppm) 0.099 0.267		
04-2150	Carbonale	11.44	2.17	1.15	3.85	39.48	<0.01	0.13	0.03	0.062		
04-2151	White saprolite	60.30	16.72	0.65	0.68	6.72	0.05	0.90	0.03	< 0.602		
04-2152	Red saprolite	43.70	12.30	32.37	0.16	0.15	0.13	0.75	<0.01	< 0.002		
04-2153	Carbonate	11.70	2.21	2.09	2.86	36.81	<0.01	0.12	0.08	0.157		
04-2154	Carbonate	18.90	3.37	4.67	3.38	32.12	<0.01	0.20	0.06	0.210		
04-2155	Red saprolite	36.97	12.93	36.66	0.48	0.37	0.11	0.70	< 0.01	0.002		
04-2156	Carbonale	18.57	3.24	3.63	3.42	33.22	0.15	0.20	0.06	0.173		
04-2157	Carbonate	17.81	3.37	2.94	3.46	33.13	0.15	0.19	0.02	0.199		
04-2158	Red saprolite	34.55	12.63	37.91	0.50	0.41	0.10	0.69	<0.01	0.012		
SAMPLE	TYPE	Ba	Ве	Со	Cr	Cu	Mn	Ni	٧	Zr		
04-2112	Whole	1089	1	2	771	82	47	52	460	71		
04-2129	Whole	686	<1	<1	417	39	120	28	579	111		
04-2150	Carbonale	889	1	13	62	35	43	17	25	25		
04-2151	White saprolite	55	2	13	42	26	3	47	32	93		
04-2152	Red saprolite	9	<1	15	397	<5	38	37	413	128		
04-2153	Carbonate	13385	<1	18	88	26	79	41	35	27		
04-2154	Carbonate	1392	<1	13	163	42	139	51	73	51		
04-2155	Red saprolite	207	<1	4	492	21	12	34	397	135		
04-2156	Carbonate	4814	3	:0	155	45	87	39	76	45		
04-2157	Carbonate	4903	1	5	129	46	83	40	47	60		
04-2158	Red saprolite	313	<1	7	592	24	21	39	458	132		
SAMPLE	TYPE	S	Се	Ga	Ge	Мо	Nb	Pb	Rb	Sr	Y	
04-2112	Whole	0.06	9	19	4	4	2	4	18	64	8	
04-2129	Whole	0.05	18	18	<1	2	3	6	1	69	5	

SAMPLE	DEPTH (m)	WEIGHT WEIGHT Minus Plus		Au Minus	A u PLUS	Au Minus	A u PLUS
	(111)	0.2mm	0.2mm	0.2mm	0.2mm	(mg)	(mg)
2113	0.025	386.3	104.3	31	3	11.98	0.31
2114	0.075	292.3	199.2	29	13	8.48	2.59
2115	0.125	302.2	193.2	18	1	5.44	0.19
2116	0.175	355.9	136.5	27	1	9.61	0.14
2117	0.250	403.1	87.7	47	2	18.95	0.18
2118	0.350	421.7	67.0	32	2	13.49	0.13
2119	0.450	400.6	85.6	52	10	20.83	0.86
2120	0.550	380.8	105.8	75	43	28.56	4.55
2121	0.650	350.1	136.5	76	31	26.61	4.23
2122	0.750	356.5	152.7	223	54	79.50	8.25
2123	0.850	406.9	86.2	160	73	65.10	6.29
2124	0.950	420.0	66.9	137	66	57.45	9.16
2125	1.100	206.2	289.9	123	15	35.66	4.35
2126	1.300	140.2	359.0	105	16	14.72	5.74
2127	1.500	286.3	199.6	39	28	11.17	5.59
2128	1.700	393.0	103.1	7	7	2.75	0.72